

1 Distribution of Te, As, Bi, Sb and Se in; MORB, Komatiites and in Picrites and Basalts
2 from Large Igneous Provinces: Implications for the Formation of Magmatic Ni-Cu-PGE
3 Deposits

4 *Authors: *Sarah-Jane Barnes¹, Eduardo T. Mansur^{1,2},*

5 ¹Sciences de la Terre, Université du Québec à Chicoutimi, Québec, G7H 2B1, Canada

6 ²Present address Norges geologiske undersøkelse, Leiv Eirikssons vei 39 7040 Trondheim,
7 Norway

- 8
9 • Corresponding author Email sjbarnes@uqac.ca

10 Published as

11 S.-J. Barnes et E. T. Mansur, "Distribution of Te, As, Bi, Sb, and Se in Mid-Ocean Ridge Basalt and
12 Komatiites and in Picrites and Basalts from Large Igneous Provinces: Implications for the
13 Formation of Magmatic Ni-Cu-Platinum Group Element Deposits," *Economic Geology*, vol.
14 117, no. 8, pp. 1919-1933, 2022. doi: 10.5382/econgeo.4887.

15
16 **Abstract**

17 In magmatic nickel-copper-platinum-group element (PGE) deposits the PGE are found both
18 in solid solution in base metal sulfides and as platinum-group minerals (PGM). Apart from
19 S the most common elements that the PGE combine with to form PGM are Te, As, Bi, Sb
20 and Sn (TABS). Whether the TABS play a role in collecting the PGE or simply partition into
21 the sulfide liquid along with the PGE and later combine with PGE when the sulfide phases
22 become saturated PGE is not currently clear. This is in part because the concentrations of
23 TABS in the magmas (picrites and basalts from large igneous province and komatiites)
24 which form these of types of deposits are not well established and hence it is not evident
25 whether the magmas contain sufficient TABS to control PGE. In order to establish the
26 concentrations of Te, As, Bi, Sb and Se (TABS+) in these rock types and to document the
27 processes affecting these concentrations we have determined TABS+ concentrations in
28 komatiites, in MORB and in picrites and basalts from large igneous provinces. Using TABS+

29 mantle normalized diagrams the affects of; different mantle sources, crystal fractionation,
30 crustal assimilation, degassing and alteration are considered. We estimate the concentrations
31 of TABS+ in komatiites to be approximately twice primitive mantle values. In picrites the
32 concentrations vary from approximately ten times primitive mantle values for As and Sb and
33 decrease through Bi to Te from seven to two times primitive mantle.

34 Assimilation of S-bearing sedimentary rocks is thought to be important in triggering sulfide
35 saturation leading to the formation of Ni-Cu-PGE deposits. Assimilation of such sediments
36 would enrich the magma in Th over Nb and in As, Sb and Bi. Evidence of assimilation is in
37 the form of TABS and Th enrichment is clear in the PGE reef deposits of the Bushveld and
38 Stillwater Complexes, but the deposits do not contain sufficient TABS to control the PGE.
39 This is also true in the Noril'sk-Talnakh Ni-Cu-PGE deposits. However, at Noril'sk
40 degassing of the magma has resulted in the loss of TABS which results in negative As, Bi,
41 Se and Te anomalies on primitive mantle normalized plots.

42 Key words: Tellurium, Arsenic, Antimony, Selenium, Bismuth, platinum-group elements,
43 magmatic ore deposits, crustal assimilation; sulfide segregation; degassing, Siberian flood
44 basalts, komatiites, MORB, Karoo, Etendeka, Emeishan.

45

46

47

Introduction

48 Tellurium, As, Bi, Sb and Sn (TABS) are essential components of most platinum-
49 group minerals (PGM) found in magmatic Ni-Cu and platinum-group-element (PGE)
50 deposits and hence they may play a role in the formation of these deposits. However, to
51 consider whether they play an active role (i.e. they act as collectors of the PGE; Hanley,
52 2007; Helmy et al., 2010, 2020; Piña et al., 2015,; Anenburg and Marvogenes, 2016; Cafagna
53 and Jugo, 2016; Liang et al., 2019;) or whether their role is passive (i.e. both TABS and
54 PGE partition from a silicate magma into a sulfide liquid and eventually the sulfide liquid or
55 sulfide minerals becomes saturated in the PGM; Dare et al., 2014; Liu and Brenan, 2015;
56 Duran et al., 2017; Mansur et al., 2020a; Mansur and Barnes, 2020a) it is necessary to
57 estimate their concentrations in the silicate magmas from which the sulfide liquid segregates.

58 Most Ni-Cu-PGE deposits are found in association with either komatiites or mafic
59 magma of large igneous provinces (Barnes and Lightfoot, 2005; Wilson, 2012; Maier et al.,
60 2013; Barnes et al., 2016; Jenkins and Mungall, 2018). However, because of analytical
61 difficulties there are very few determinations of TABS concentrations in these rock types
62 (Mansur et al., 2020b). Therefore, we have determined the concentrations of Te, As, Bi Sb
63 and Se (hereafter abbreviated to TABS+ to include Se) from 116 samples of; picrites and
64 basalts from large igneous provinces (LIPs), komatiite and MORB. These samples were
65 selected because they have been previously studied and whole rock major and trace element
66 (including S, Cu and PGE) contents were already known.

67

68

69

In this work we show that establishing the initial TABS+ contents of magma that
formed these rock types is complex as these elements are affected by numerous competing
processes. The MORB lavas show that segregation of sulfide liquid leads to strong depletion

70 of Te, Pd and Pt relative to the other TABS+ and Cu. Contamination with continental crust
71 enriches the magma in Th, As, Sb and Bi relative to Nb, Cu, Se, Te, Pd and Pt. Degassing
72 leads to depletion of As, Bi Se and Te relative to Th, Nb, Cu, Sb, Pd and Pt. Alteration leads
73 to enrichment in As, Sb and Bi. (In this work we use the word alteration to refer to changes
74 in rock composition brought about by either metamorphism or hydrothermalism). In order
75 to separate these processes, we have developed primitive mantle normalized plots on which
76 the elements are plotted in order of partition coefficient between silicate and sulfide liquid.
77 We then apply these results to modeling the concentrations of TABS+ in the Bushveld and
78 Stillwater PGE deposits and to the Ni-Cu-PGE deposits of Noril'sk-Talnalk district.

79 **Materials and Methods**

80 *Materials*

81 The samples are: Komatiites from the Abitibi and Belleterre (Canada) and Barberton
82 (South Africa) Greenstone belts; basalts from the Cape Smith fold belt (Canada); MORB
83 from, the Hotu seamount, the Garret fracture zone and the South Atlantic Ridge; picrites and
84 basalts from the Siberian, Emeishan, Etendeka and Karoo LIPs (Fig. 1.) The samples were
85 chosen because they had been previously investigated to understand their PGE content. More
86 details on the samples can be found in electronic supplementary materials 1 (App.1).

87

88 *Analytical Methods*

89 Tellurium, As, Bi, Sb and Se analyses were carried out by Hydride Generation-
90 Atomic Fluorescence Spectrometry (HG-AFS) following the technique described by Mansur
91 et al. (2020b), at LabMaTer Université du Québec à Chicoutimi (UQAC). International
92 reference materials (CH-4 and TDB-1 from Natural Resources Canada and OKUM from
93 IAGEO), and a blank were determined at the same time as the samples. The detection limits
94 based on 3σ of the blank are 0.005, 0.003, 0.005, 0.005 and 0.002 ppm for Te, As, Bi, Sb

95 and Se, respectively. The results for the reference materials agree with working values (App
96 2 Table A1).

97 For most samples, whole-rock analyses were already available in the original
98 publications. For samples where they were not, the samples were analyzed at LabMaTer,
99 (UQAC) by LA-ICP-MS details in Appendix 1 The certified reference materials OKUM,
100 KPT-1 (IAG reference materials), WPR-1 (CANMET) and UB-N (CNRS-CRPG), were
101 used to monitor the results. The results obtained for the reference materials agree with the
102 working values (App 2, Table A2).

103 *Rational for the use of primitive mantle normalized plots*

104 The data is presented on primitive mantle normalized diagrams. The mantle
105 normalization factors used were those of Lyubetskaya and Korenaga (2007). Wang et al.
106 (2018) provide more recent estimations based on a combination of Lyubetskaya and
107 Korenaga (2007), McDonough and Arevalo (2008) and Palme and O'Neil (2014) and a
108 statistical approach to minimize errors. Estimations of the concentrations of TABS+ in the
109 primitive mantle are poorly constrained because of the many processes affecting their
110 distribution and analytical difficulties. The concentrations of As, Sb and Bi in these
111 publications are based on the ratios of these elements to Pb or Ce in ocean floor basalts and
112 concentrations of Se and Te are based on the ratio of these elements to S in CI chondrites.
113 Wang's et al. (2018) estimations are within error the same as those of Lyubetskaya and
114 Korenaga (2007) for the elements considered here and the shape and level of the patterns
115 presented would not significantly change if Wang et al.'s (2018) values were used.

116 In the crust and mantle the TABS+ behave as chalcophile elements (i.e. they partition
117 into base metal sulfide liquid and minerals; Hattori et al., 2002; Patten et al., 2013; Brennan,
118 2015; Li and Audéat, 2015; Liu and Brennan, 2015; Greaney et al., 2017). Therefore, during

119 partial melting of the mantle the presence of sulfides in the restite may deplete the magma
120 in TABS+, and the segregation of a sulfide liquid during transport of crystallization will have
121 a similar effect. However, the TABS+ are not equally chalcophile. Arsenic and Sb are
122 slightly chalcophile ($D^{\text{Sulf liq/Sil liq}} < 10$), Bi, Se and Cu are strongly chalcophile ($D^{\text{Sulf liq/Sil liq}}$
123 from 100 to 1000) and Te is highly chalcophile ($D^{\text{Sulf liq/Sil liq}} > 1000$) (Barnes, 2016 and
124 references therein). Thus, the TABS+ may be fractionated from each other by segregation
125 of a sulfide liquid. In order to assess the effect of sulfide liquid segregation or sulfide liquid
126 accumulation the elements are arranged in order of partition coefficient into a sulfide liquid
127 (Fig. 2A).

128 Sulfides are not always present during partial melting and crystallization, which is
129 important because the TABS+ are incompatible with most mafic minerals (Kamenetsky and
130 Eggins, 2012; Jenner and O'Neill, 2012; Maciag and Brenan, 2020; Mansur and Barnes,
131 2020b) and thus should correlate with lithophile incompatible elements. To take into
132 consideration the effects of degree of partial melting and/or crystal fractionation Th, Nb (or
133 Ta) were added to the diagram as representatives of the lithophile incompatible elements
134 (Fig. 2A)

135 Contamination of mafic magmas by continental crust is common, and the continental
136 crust, and in particular black shale, is enriched in As, Sb and Bi (Fig. 2B; Ketris and
137 Yudovich, 2009; Godel et al., 2012; Piña et al., 2013, 2015; Samalens et al., 2017; Le Vaillant
138 et al., 2018). Thus, contamination with continental crust results in enrichment in As, Sb and
139 Bi as has been previously shown for the Bushveld Complex B-1 dikes (Mansur and Barnes,
140 2020b Fig. 2B). Thorium is also enriched in continental crust rocks, thus if a magma has
141 experienced contamination by continental crust a negative Nb or Ta anomaly is observed
142 (Fig. 2B).

143 The TABS+ and S are volatile (Lodders, 2003; Wood et al. 2019), and therefore
144 subaerial lavas could be depleted in these elements during degassing (Iacono-Marziano et
145 al., 2012, 2017; Zelensky et al., 2014; Mather et al., 2015; Forrest et al., 2017; Edmonds and
146 Mather, 2017; Edmonds et al., 2018; Cox et al., 2019; Wiesener et al., 2020). In order to take
147 this into consideration, chalcophile elements that are less volatile (Cu, Pd, and Pt) were
148 added to the plot at the position of their relative partition coefficients into base metal sulfide
149 liquid. If degassing has occurred one would predict that there should be depletion of the
150 TABS+ relative to Cu, Pd and Pt (Fig. 2A).

151 Finally, it is well established that the TABS+ can be mobilized either during
152 metamorphism (Pitcairn et al., 2015; Hammerli et al., 2016) or hydrothermal activity (Guo
153 and Audétat, 2017; Patten et al., 2017; Stucker et al., 2017; Shevko et al 2018;). Deviations
154 from the trends shown by the immobile elements may be the result of alteration.

155 **Results and Interpretation**

156 The overall geochemical characteristics of samples from the various localities are
157 presented using primitive mantle normalized plots (Fig. 3). Median values for each locality
158 are presented in (Table 1). The plots for individual samples are shown in the Appendix 3
159 (Fig. A1 and A2). The individual whole-rock results can be found in Table A3.

160 *MORB*

161 The MORB samples show two slightly different primitive mantle normalized
162 patterns (Fig. 3A). The patterns for samples from the South Atlantic Ridge and the Garret
163 Fracture Zone display a typical N-MORB shape, with relatively flat slopes from Th to Se at
164 approximately 1-10 times primitive mantle, followed by steep negative slopes from 0.5-2
165 times primitive mantle at Te to 0.001-0.1 times primitive mantle at Pt. The E-MORB samples
166 have higher Th and Nb contents at 10 to 40 times primitive mantle but have similar pattern

167 to N-MORB samples for the chalcophile elements (Fig. 3A). Jenner and O'Neill (2012)
168 reported similar results for As, Bi, Sb, Cu and Se for a large dataset of MORB glasses,
169 whereas Yi et al. (2000) reported results for Te which are slightly lower than ours (0.001 to
170 0.01 ppm vs <0.005 to 0.014 ppm).

171 Patten et al. (2013) reported on the presence of sulfide droplets in these samples and
172 modelled the strong Pt and Pd depletion by the segregation of an immiscible sulfide liquid
173 during crystal fractionation. However, the relatively flat primitive mantle normalized
174 patterns of the N-MORB samples from Th to Se do not show evidence of sulfide segregation.
175 This is because, it is the bulk partition coefficients ($D^{\text{sulf liq/sil liq}} * \text{weight fraction of sulfide}$
176 liquid segregated) that controls the behavior of the elements. If the MORB samples
177 originally contained Pd and Pt at the same levels as Th to Se, (~3 times primitive mantle),
178 only a very small amount of sulfide liquid needs to have segregated to lower the
179 concentrations of Te through to Pt to their current levels. Assuming saturation of the basalts
180 at 1000 ppm S and allowing for ten percent equilibrium fractionation with cotectic
181 segregation of the sulfide liquid, the Pd and Pt concentrations would be lowered to the
182 median levels seen in our MORB samples. The concentrations of Te with its high partition
183 coefficients into sulfide liquid ($D^{\text{sulf liq/sil liq}} > 5000$; Patten et al., 2013; Liu and Brenan, 2015)
184 would be similarly affected. In contrast, because of the lower partition coefficients of the
185 other chalcophile elements they are only slightly affected (model line on Fig. 3A).

186 Here we have modelled the depletion of Te, Pd and Pt as though it occurred in the
187 oceanic crust because the samples contain sulfide droplets and were sulfide saturated. If, on
188 the other hand, a small amount sulfide remained in the mantle source during partial melting
189 the result would be the similar.

190 The incompatible elements Th and Nb are markedly enriched in E-MORB (Fig. 3A)
191 relative to the chalcophile elements. This suggests that whereas an E-MORB source is
192 enriched in incompatible lithophile elements it is not enriched in chalcophile elements and
193 that this enrichment is not due in continental crustal component as both Th and Nb are
194 enriched.

195 *Cape Smith*

196 The Cape Smith basalts have an almost flat pattern at 2 to 6 times primitive mantle.
197 Antimony and Bi are slightly enriched relative to the other elements. The levels of Th, Nb,
198 Bi, Cu, Se are similar to N-MORB, but the levels of the highly chalcophile elements Te, Pd
199 and Pt are much higher than MORB (Fig. 3B). These basalts have been modeled as the
200 product of approximately 20% partial melting of the mantle and very little retention of
201 sulfide minerals in the mantle (Barnes and Picard, 1993). Assuming a primitive mantle
202 source the TABS primitive mantle normalized pattern in this case should be essentially flat
203 at 5 times primitive mantle (Fig. 3B) with no depletion in Te, Pd and Pt. The median TABS
204 patterns match the model pattern except for the enrichment of Sb and Bi. This enrichment
205 could be the product of seafloor alteration, as these elements have been shown to be mobile
206 during seafloor alteration (Patten et al., 2017). But it could also have occurred during
207 metamorphism as the rocks have experienced greenschist facies metamorphism (Picard et
208 al., 1990). Pitcairn et al. (2015) showed that As and Sb can be enriched in basalts that have
209 undergone lower greenschist facies metamorphism (Bi was not included in their study).

210 *Komatiites*

211 The Al-undepleted komatiites from the Abitibi and Belleterre Greenstone belts
212 display similar patterns, which, with the exceptions of As, Sb and Bi, are relatively flat at 1

213 to 6 times primitive mantle. The samples have positive As, Sb and slight Bi positive
214 anomalies at 8 to 30 times primitive mantle (Fig. 3C).

215 Based on the depletion of highly incompatible elements (La, Zr and Hf), which show
216 a depletion factor of ~ 0.6 relative to moderately incompatible elements, Barnes et al. (1983)
217 modelled the Alexo komatiites as the product of 25 to 40% partial melt of depleted mantle.
218 The concentrations of Th and Nb at one times primitive mantle are consistent with this model
219 (Fig. 3C). The level of Cu through to Pt at approximately two times primitive mantle is
220 similar to that of primitive mantle for the moderately incompatible elements (Sm, HREE and
221 TiO_2) for these rocks. We interpret this to be because the elements Cu to Pd behaved as
222 moderately incompatible elements during the earlier partial melting event which initially
223 depleted the komatiite source in highly incompatible elements.

224 The Al-depleted komatiite samples from the Barberton Greenstone Belt also show
225 enrichment in As, Sb and Bi. The Al-depleted komatiites but are relatively depleted in the
226 other chalcophile elements which are mostly present at less than one time primitive mantle
227 (Fig. 3D). The Hooggenoeg Formation samples are enriched in Th relative to Nb (Fig. 3D).

228 The depletion of Cu, Se, Te, Pd and Pt could be due to segregation of a sulfide liquid.
229 However, Maier et al. (2009) argued that the low concentrations of PGE in these rocks and
230 other komatiites of the early Archean is due to the mantle source being depleted in PGE due
231 to incomplete mixing of the late veneer. One of three formations (the Hooggenoeg) is
232 enriched in Th over Nb suggesting that the Hooggenoeg samples have experienced crustal
233 contamination.

234 All of the komatiites are enriched in As, Sb, and Bi relative to the other elements at
235 10 to 20 times primitive mantle (Fig. 3C and D). This enrichment is not thought to be due to
236 crustal contamination because Th is not enriched relative to Nb (except in the Hooggenoeg

237 Formation). Neither is the enrichment thought to be due to enrichment in the komatiite
238 mantle source because these elements do not behave in a coherent fashion. For example, at
239 Alexo (where the komatiites showing the lowest degree of metamorphism and alteration of
240 the komatiites) a plot of Nb versus Mg# shows a typical trend for incompatible element with
241 Nb increasing as Mg# drops, and the spinifex-textured and chill samples (+) having higher
242 Nb values than the olivine-rich (o) lower parts of the flows (Fig. 4A). In contrast, Sb, As and
243 Bi concentrations do not correlate with Mg# (Bi distribution shown in Fig. 4B). As
244 mentioned above basalts that have undergone greenschist metamorphism are enriched in As
245 and Sb (Pitcairn et al. 2015) thus the enrichment of As, Sb and by analogy Bi in the
246 komatiites could be the result of metamorphism.

247 *Large Igneous Provinces*

248 Most of the samples from LIPS show similar patterns with overall negative slopes
249 decreasing from Th and Nb (primitive mantle normalized values in the 10 to 70 range) to Pd
250 and Pt values in the 1-3 range (Fig. 3E to H). The Nd lavas from the Siberian province are
251 an exception containing even lower values of Pd and Pt. Most samples from LIPs also show
252 negative As, Bi, Se and Te anomalies (Fig. 3E to H), with the exceptions that Dali picrites
253 and the Etendeka dikes which do not show Bi anomalies (Fig. 3E and F).

254 The patterns from different localities, do however, show differences in Th to Nb
255 ratios. Most of the Emeishan samples, the picrite dikes from Etendeka and the picrite from
256 Tuli (Karoo) do not show Th enrichment over Nb, (Figs. 3E, F and G). In contrast, rocks
257 from almost all of the other Karoo formations and the Siberian lavas are enriched in Th (Fig.
258 3G and H).

259 The patterns of the LIPs rocks are more complex than those of MORB and komatiites
260 where a combination of differences in mantle source, sulfide segregation and alteration were

261 sufficient to explain the patterns. In the case of LIPs degassing and continental crustal
262 contamination must be considered. These samples were not metamorphosed or
263 hydrothermally altered so alteration not as important as in the case of the komatiites.

264 *Degassing* _The most striking difference between the sub-aqueous rocks (MORB, Cape
265 Smith and komatiites) and LIPs rocks are the negative As, Bi, Se, and Te anomalies. We
266 attribute these anomalies to magma degassing for the following reasons: i) The low Se and
267 Te values cannot be attributed to sulfide segregation because Se has similar partition
268 coefficient to Cu into sulfide liquid and thus would be expected to be present at similar
269 levels to Cu. Similarly, Te has a slightly lower partition coefficient into sulfide liquid than
270 Pd and thus would be expected to be present at a higher or similar level to Pd; ii) The
271 anomalies are not due to continental crust contamination because as will be discussed
272 below this would enrich the magmas in As, Sb and Bi rather than depleting them; iii) The
273 lavas contain vesicles indicating gas loss; iv) Sub-aqueous mantle derived magmas
274 generally contain 1000 ppm S whereas most of our LIPs rocks contain <100 ppm
275 indicating S loss; and v) Like S, the TABS+ are volatile (Lodders, 2003; Wood et al.,
276 2019) thus are prone to loss during degassing.

277 Zelenski et al. (2014) report partition between gas and basalts from Erta Ale,
278 Tolbachik and Kudryavy volcanoes in the 2 to 1000 range for As, Bi, Se, and Te. For Cu,
279 Sb and Pt the partition coefficients are in the 0.001 to 0.1 range (Fig. 2A). Zelenski et al.
280 (2014) do not report partition coefficients for Th or Nb, however given their low volatility
281 (Loedders, 2003; Wood et al., 2019) they would not be expected to partition into a gas. The
282 difference in the partition coefficients between gas and silicate liquid for As, Bi, Se, Te and
283 Sb, Cu and Pt could explain the negative anomalies for As, Bi, Se and Te. These observations
284 should be tempered with consideration that in some studies Cu, Sb and Pt have been found
285 to be moderately volatile (Mather, 2015). The differences in behavior reported by Zelenski

286 et al. (2014) and Mather (2015) could in part be the result of differences in the oxidation
287 state of the magma (the TABS+ can adopt a range of oxidation states, (Pokrovski et al.,
288 2013)) which could affect their affinity with the gas versus the magma. It is also important
289 to consider when degassing occurred relative to the phases present in the magma (Edmonds
290 and Mather, 2017). Of particular importance is the presence of sulfide liquid and aqueous
291 fluids into which the TABS could also preferentially partition and thus not partition into the
292 gas. In any event, all the TABS+ except Sb appear to have been lost from the LIPs samples.

293 Continental Crust Contamination. Contamination of the magmas with continental crust or
294 sediment would enrich the samples in As, Sb and Bi (Fig. 2B and 3H). This enrichment
295 could potentially obscure the effects of degassing. For example, consider the Etendeka
296 picrite with a flat pattern but negative As, Se and Te anomalies (Fig. 5A). Contamination
297 with ~ 20% upper continental crust accompanied by 60% crystal fractionation produces a
298 pattern that matches the most evolved Etendeka dolerite sample. The dolerite does not
299 show negative As, Bi or Te anomalies because the continental crust has added sufficient of
300 these elements to eliminate the anomalies. However, there is still a negative Se anomaly
301 because the continental crust does not contain much Se (Hu and Gao, 2008).

302 Most of the Karoo basalts and dolerites are enriched in Th relative to Nb, which is in
303 agreement with previous work (Marsh et al., 1997) that showed that lavas from these areas
304 have assimilated upper continental crust material. The Tuli basalts can be modeled by 10%
305 crustal contamination of the Tuli picrite (Fig. 5B). The Lesotho and Barkley East basalts
306 have Nb and Th contents lower than the Tuli picrite making this picrite an unsuitable parental
307 magma for these basalts. A magma similar to the Etendeka picrite would be a suitable
308 parental magma. Ten percent crustal contamination of the Etendeka picrite accompanied by
309 20% crystal fractionation would produce a TABS+ pattern similar to the Lesotho and
310 Barkley East Karoo basalts (model line in Fig. 3G), except that the Karoo basalts show a

311 negative Bi anomaly. Possibly, the Karoo basalts have experienced greater degassing than
312 the Etendeka rocks, which are dikes.

313 As in the case of the Karoo basalts the Siberian basalts are all enriched in Th over
314 Nb which is in agreement with Lightfoot et al. (1993) that they have experienced continental
315 crust contamination. The Morongovsky (Mr), Mokulaevsky (Mk) and Kharaelakhsky (Kh)
316 formations are similar to the Lesotho and Barkley East Karoo basalts and can be modelled
317 by in a similar manner by contamination with approximately 5% upper continental crust
318 accompanied by 7% crystal fractionation (Fig. 3H). The Nd lavas are more enriched in Th
319 relative to Nb and require a greater degree of contamination (26 %) accompanied by a greater
320 amount of crystal fractionation (30%). More strikingly they show strong depletion from Cu
321 to Pt. These lavas are believed to have been depleted in chalcophile elements due to the
322 segregation of sulfide liquid (Lightfoot et al., 1993; Brüggmann et al., 1993; Lightfoot and
323 Keays, 2005). Assuming that the lavas originally contained a similar amount of Pd and Pt to
324 the Mr/Kh lavas only a small amount of sulfide liquid would need to have segregated to
325 lower the Pd and Pt values. If cotectic proportions of sulfides segregated in the last 3% of
326 crystal fractionation the Pd and Pt concentrations would have been reduced to Nd levels (Fig.
327 3H).

328 **Discussion**

329 *The Nature of the Mantle Sources*

330 The Al-undepleted komatiites and Cape Smith basalts contain Se, Te, Pd, and Pt at
331 approximately slightly higher levels than Th and Nb for the komatiites and similar levels for
332 the Cape Smith basalts. We interpret this to imply a slightly depleted mantle source for the
333 komatiite and an undepleted mantle source for the Cape Smith Basalts with no sulfide phase
334 retention. The N-MORB contain Th, Nb, As, Sb, Bi, Cu and Se at approximately similar

335 primitive mantle levels. Given that the N-MORB mantle is depleted in highly incompatible
336 elements we interpret this to indicate that degree of As, Sb, Bi Cu and Se depletion in the N-
337 MORB mantle is similar to that of highly incompatible elements.

338 All of the LIPs picrites are enriched in the highly incompatible elements Th and Nb
339 relative to the chalcophile elements (Fig. 6). The exact nature of the mantle from which the
340 magmas of LIPs form is a much-debated topic with views ranging from formation by partial
341 melting of a source consisting of primitive mantle, MORB and sediment (Zhang et al., 2019)
342 to formation by partial melting of metasomatized sub-continental lithosphere (Kamenetsky
343 et al., 2012; Shellnutt, 2013). Most of the picrites, (Fig. 6) do not show enrichment in Th
344 over Nb and the TABS+ (the Binchuan picrites are an exception showing slight Th
345 enrichment). The picrites TABS+ patterns however are not all the same and can be divided
346 into two groups based on their Th and Nb contents. The Dali and Etendeka picrites show
347 much lower Th and Nb contents at 10 times primitive mantle than the other Emeishan picrites
348 and the Tuli picrite (Karoo), which contain approximately 30 times primitive mantle levels
349 of Th and Nb indicating a much more enriched source for these picrites (Fig. 6). As in the
350 case of N-MORB vs E-MORB this enrichment in Nb and Th in the Tuli and Emeishan
351 picrites vs the Etendeka and Dali picrites, does not apply to the chalcophile elements (Figs.
352 3A and 6). This suggests that the process that enriched the sources of E-MORB and many
353 LIPs picrites in Th and Nb did not enrich them in chalcophile elements. Furthermore, the
354 component leading to the enrichment of Th and Nb is not a subducted sediment, as Th is not
355 enriched over Nb.

356 As pointed out by Zhang et al. (2019) the high Mg# of the Emeishan picrites requires
357 a high degree of melting of a peridotite source. A high degree of partial melting of a
358 peridotite source is consistent with the Pd and Pt concentrations observed in the picrites at 1
359 to 2 times primitive mantle. On the other hand, the high concentrations of Nb at 10 to 30

360 times primitive mantle, Sb and Cu primitive mantle normalized levels at 5 and 10 times
361 mantle would require a low degree of partial melting if the mantle source was similar to
362 primitive mantle. The added component that enriched the mantle source must then: i) Not
363 dilute the MgO content of the source, in order to keep the high Mg# of the picrites, therefore
364 it must be a small volume; ii) Be very rich in **both** Th and Nb, i.e. not a sediment or melt of
365 a sediment; iii) Contribute some Cu and Sb; and iv) Probably contribute some As, Bi, Se,
366 Te, however, as these elements are partially lost in subsequent degassing this is not a firm
367 requirement. Exactly what is the nature of this component is not clear to us, but possibly a
368 carbonate melt or fluid might be suitable (e.g. Holwell et al., 2019; Blanks et al., 2020).

369 *Estimation of TABS+ concentrations in komatiite and LIPs magmas*

370 The motivation for this study was to estimate the initial TABS+ concentrations in
371 komatiite and LIPs magmas because these types of magmas are thought to host magmatic
372 Ni-Cu-PGE deposits. Simply calculating the average of TABS+ values for komatiites and
373 LIPS rocks is not sufficient to estimate these concentrations because of the various processes
374 that have modified them. The reasoning for the choice of values are proposed here discussed
375 as below.

376 The Al-undepleted komatiites and Cape Smith basalts show similar primitive mantle
377 normalized concentrations for Cu, Se, Te, Pd and Pt at approximately 2 and 5 times primitive
378 mantle, respectively, consistent with sufficiently high degrees of partial melting (40% and
379 20%, respectively) to dissolve all the sulfide minerals in the source and release these
380 elements into the melt. Based on these observations the concentrations of Se and Te in Al-
381 undepleted komatiite can be estimated as approximately twice primitive mantle (Table 2).

382 Bismuth and Sb are present at higher levels than Th, Nb and the other chalcophile
383 elements in the komatiites and the Cape Smith basalts. This is attributed to metamorphism

384 of the samples and hence the concentrations of Bi and Sb cannot be empirically estimated.
385 Similarly, As in the komatiites is higher than Th, Nb and other chalcophile elements and this
386 could be due to enrichment during metamorphism. However, As is present in the Cape Smith
387 basalts at the same level as Nb and Cu, at five times primitive mantle, and could be primary.

388 Our N-MORB samples have a median Mg# of 0.65 and hence are close to primary
389 magmas. Thorium, As, Sb, Bi, Cu and Se are all present at approximately 3 times primitive
390 mantle levels and provide a rationale for estimating the As, Sb and Bi concentrations in
391 komatiites by assuming that they are present at the same primitive mantle levels as Th, Cu
392 and Se, i.e. twice primitive mantle (Table 2).

393 Most LIPs picrites appear to be depleted in TABS+ (except Sb) relative to the less
394 volatile elements on the primitive mantle normalized plots. We attribute this depletion to
395 degassing. To estimate the concentrations of the TABS+ before degassing the primitive
396 mantle ratios of Cu/Bi, Cu/Se and Pd/Te could be used. (Assuming Cu and Pd were not
397 significantly depleted during degassing). Based on the similarity of TABS primitive mantle
398 normalized concentrations in N-MORB and for Te and Se in komatiites and the Cape Smith
399 basalts, the original concentrations in the picrites can be estimated by using the adjacent
400 elements. Copper is present at approximately 5 times primitive mantle allowing an estimate
401 for Bi and Se of 5 times primitive mantle and Pd is present at twice primitive mantle allowing
402 an estimate for Te at twice primitive mantle (Table 2).

403 In the Dali and Etendeka picrites the Th, Nb, Sb levels are approximately the same
404 at 10 times primitive mantle, which implies that the As content should also be 10 times
405 primitive mantle. Estimating the As content for the other picrites is more difficult because
406 Th and Nb are markedly enriched at approximately 30 times primitive mantle, whereas Sb
407 is present at approximately 10 times primitive mantle, as observed for the Dali and Etendeka
408 picrites. In E-MORB, As is present at the same primitive mantle level as Sb, Bi and Cu at 3

409 times primitive mantle rather than the more than 10 times primitive mantle observed for Th
410 and Nb. Therefore, to estimate the concentration of As in the picrites we have used the
411 primitive mantle concentration of Sb at 10 times primitive mantle (Table 2).

412

413 *Application to Ni-Cu-PGE Deposits*

414 *PGE Deposits* The main PGE ore deposits are the Merensky reef, the Platereef, the UG2
415 reef (all of the Bushveld Complex), the JM reef of the Stillwater Complex and the Main
416 Sulfide zone of the Great Dyke (Naldrett, 2011; Zientek, 2012). Very limited data is
417 available for the UG2 reef and the Main Sulfide zone and they will not be considered here.
418 The primitive mantle normalized plot for the normal and thin reef from the Merensky and
419 JM reef both show an enrichment of Th (at three to ten times primitive mantle) over Nb and
420 Ta (at 0.1 to 1 times primitive mantle) (Figs. 7A and C). (Data from Mansur and Barnes
421 2020). This is consistent with the magmas having been contaminated with continental
422 crust. The levels of the patterns increase steadily from approximately one times mantle at
423 As through to approximately 100 times primitive mantle at Bi. The patterns show a slight
424 dip through Cu to Te and then increase markedly to 1000 to 10 000 times primitive mantle
425 (Figs. 7A and C).

426 The increase in chalcophile element contents from As through to Pt is consistent with
427 the increase in partition coefficients between sulfide and silicate liquid (Liu and Brenan,
428 2015). The slight Bi anomaly is the product of the much higher partition coefficient for Bi
429 than for Sb between sulfide and silicate liquid and the enrichment of Bi over Cu in the
430 magma due to the contamination of the magma with continental crust (Fig. 2B). The
431 distribution of the chalcophile element content of the normal and thin Merensky reef at
432 Impala and Rustenburg mines have been modelled in detail using a magma of similar
433 composition to the marginal chills of the Bushveld (Mansur and Barnes, 2020b). On average,

434 a model consisting of 2 weight percent sulfide liquid formed at an R-factor of 30 000 and 20
435 percent trapped liquid component models the composition of these rocks (Fig. 7A). Thus the
436 PGE content of the Rustenburg and Impala normal and thin reefs can be modelled based by
437 collection of PGE and TABS from a silicate magma by sulfide liquid and do not require
438 TABS to collect the PGE.

439 In addition to the previously published data for the Merensky reef a composite
440 sample, SARM-7, prepared by National Institute of Metallurgy South Africa from 7.5 tonnes
441 of Merensky reef from 4 of the mines of the western limb of the Bushveld, (Steele et al.,
442 1975) was also analyzed. The pattern for a composite sample from the Merensky reef is
443 slightly different to the Impala and Rustenburg normal and thin reef in that it is an order of
444 magnitude richer in both As and Sb (Fig. 7B). This is not an exception. Scanning the results
445 reported for reference materials by African Mineral Standards for Merensky reef samples
446 shows that these are also enriched in As and Sb (Fig. 7B). All of these samples are described
447 on their certificate of analyses as composite samples from Anglo American mines from the
448 western limb of the Bushveld Complex. Furthermore, reference material from African
449 Mineral Standard for the samples from the Mogalakwena mine of the Platreef also show As
450 and Sb enrichment (Fig. 7B).

451 In the case of the Merensky reef it is worth considering that in addition to normal and
452 thin reef there are a number of other facies; contact reef, pothole reef, rolling reef, wide reef
453 (Viljoen, 1999). The pothole reef in particular is thought to be associated with the migration
454 of fluids (Kinloch, 1982; Boudreau, 1992; Viljoen, 1999). It is possible that composite
455 samples included some of the pothole reef facies and that the high Sb and As were introduced
456 by the migrating fluids. Against this is the observation that Roberts et al. (2007) showed that
457 the potholes have similar whole rock compositions to normal reef for most elements
458 (although they did not determine the TABS+ concentrations). Roberts et al. (2007) argued

459 that the potholes are product of magmatic slumping and that fluids have not substantially
460 altered the composition of the potholed reef. The high As and Sb concentrations in the
461 composite samples and particularly in the case of the Platreef may be the product of the
462 magma having been locally contaminated with rocks such as a black shale, which could have
463 contributed more As and Sb to the magma than observed in the average chill samples used
464 for the modeling (Fig. 2A).

465 TABS+ analyses of the chilled margins of the Stillwater are not available. However,
466 it is thought that the magmas from which the Stillwater formed are similar to the Bushveld
467 complex (Barnes et al., 2020). The modeled values of the chalcophile elements from the
468 Merensky reef are shown for comparison on Fig. 7B. The model composition of the
469 Merensky reef is similar to that of the JM reef for most chalcophile elements. However, as
470 is well known Pd is far more enriched in the JM reef than the Merensky reef.

471 A number of reasons have been proposed for the high Pd content. Most recently,
472 Jenkins et al. (2020) proposed that the sulfides formed in equilibrium with a more
473 fractionated magma at very high R-factors (50 000 to 500 000). The increase in R-factor
474 could account for the higher Pd concentrations in the JM reef while not affecting the
475 concentrations of most of the elements because the partition coefficients of most of the
476 chalcophile elements are less than 10 times the R-factor (Campbell and Barnes, 1984). In
477 order to explain the observation that Pt is not as enriched as Pd (as would be expected given
478 its high partition coefficient between silicate and sulfide liquid) they argue that fO_2 was low
479 (FMQ-1.5) and thus Pt had crystallized from the magma by the time sulfide saturation was
480 achieved. One difficulty with this model is the large quantity of magma required, where and
481 how the magma and sulfide liquid interacted is not considered in Jenkins et al. (2020). Barnes
482 et al. (2020) considered a number of models to account for the high Pd and high R-factor
483 and concluded on balance that the magma became contaminated with a continental crustal

484 component either at the margins of the intrusion or at depth and that the sulfides interacted
485 with a large volume of magma either as it slumped into the magma chamber from the margins
486 or if sulfide saturation occurred at depth as the magma was transported into the chamber
487 resulting in upgrading of the sulfides during transport.

488 The main alternative model favored by Boudreau (2016) is that late magmatic fluids
489 partial dissolved disseminated magmatic sulfides in the cumulate rocks underlying the reef.
490 The more soluble chalcophile elements, and in particular Pd, dissolved in the fluid. The fluid
491 rose until it reached fluid understatured magma where it dissolved into the magma. The S
492 and chalcophile elements transported by the fluid precipitated as base metal sulfide and form
493 the reef. Relevant experimental data to model this process for the TABS+ is not available.

494

495 *Noril'sk-Talnakh Nickel-Copper Deposits*_The Noril'sk-Talnakh ore camp of Siberia is the
496 largest or second largest magmatic Ni-Cu deposit in the world (Naldrett, 2011). The
497 massive sulfide ores are mineralogical and compositionally zoned (Distler, 1994;
498 Sluzhenikin et al., 2014). Duran et al. (2017) classified the sample set of massive ores
499 considered here into: i) Cu-poor (chalcopyrite cubanite-poor <10%; with pyrrhotite>70%);
500 ii) Cu-rich (chalcopyrite-cubanite-rich >50%, with pyrrhotite <10%); and iii) transitional
501 ores for those that contained intermediate amounts of pyrrhotite and chalcopyrite.
502 Pentlandite concentrations in the ore types are relatively constant at 10 to 30%. The
503 zonation is thought to be the product of crystal fractionation of the sulfide liquid with the
504 Cu-poor ores representing the early formed cumulates and Cu-rich ores forming from the
505 fractionated liquid (Distler, 1994; Duran et al., 2017).

506 The ores are found associated with three intrusions, Noril'sk I, Talnakh and
507 Kharaelakh. The primitive mantle-normalized patterns for all ore types are similar (Fig. 8).

508 They show a steep increase from As in the 10 times mantle range through to Cu in the 1000
509 (Cu-poor ore) to 10 000 times mantle range (Cu-rich ore) followed by negative Se and Te
510 anomalies, both elements being present at five to ten times less than Cu primitive mantle
511 levels (Figs. 8A to D). In the ores from Kharaelakh and Talnakh Pd is present at
512 approximately the same level as Cu, whereas in the Noril'sk I ores Pd is enriched by almost
513 an order of magnitude. Mantle normalized levels of Pt at all localities are approximately five
514 times lower than Pd.

515 All of the TABS+, are strongly incompatible with the first mineral (monosulfide solid
516 solution, MSS) to crystallize from the sulfide liquid, an exception to this is Se which is only
517 slightly incompatible (Helmy et al., 2010; Lui and Brenan 2015; Sinyakova et al., 2017).
518 Thus, although the primitive mantle normalized patterns may vary in level, the shape of the
519 patterns should be similar throughout the ore types, reflecting the trapped liquid component.
520 This can be observed for the ores associated with the Talnakh and Kharaelakh intrusions,
521 with the Cu-poor ores having the lowest the levels of TABS+ and the Cu-rich ores having
522 the highest levels, and transitional ores having intermediate levels (Figs. 8A to C). For the
523 Noril'sk I ores, TABS+ are only available for the transitional and Cu-rich ores (Fig. 8D).
524 However, the same observations can be made that the median patterns for Cu-rich ore and
525 transitional ore are similar, and the Cu-rich ore is enriched in TABS+ relative to the
526 transitional ore.

527 Assuming that the silicate magma from which the sulfides segregated was similar in
528 composition to the Mr/Kh lavas the composition of the sulfide liquid at Kharaelakh and
529 Talnakh can be modelled using an R-factor of 1 000 (consistent with previous estimates used
530 to model the PGE and Cu contents of the rocks; Duran et al., 2017). The primitive mantle-
531 normalized pattern of the model approximates the shape of the patterns for the ores except
532 the model does not show as strong a depletion in Pt relative to Pd. Platinum is known to

533 crystallize under reducing conditions (Canali et al., 2017) and is possible that the magma
534 from which the sulfides segregated had crystallized some Pt prior to sulfide segregation.

535 The model reproduces the negative Se and Te anomalies found in all ore types. Given
536 that the concentrations of these elements in the lavas appeared to be depleted due to
537 degassing we suggest that the magma from which the sulfides segregated was also depleted
538 in Se and Te due to degassing. Iacono-Marziano et al. (2012, 2017) in their studies of the
539 Noril'sk-Talnakh ores considered that degassing occurred during the formation of Norilsk-
540 Talnakh ores as consequent of crustal contamination and in fact the process was common
541 throughout the Siberian LIP.

542 The Noril'sk I ores have been modelled with higher R-factors. Using an R-factor of
543 10 000 the TABS+ primitive mantle pattern approximates the observed patterns. In fact,
544 because most of the TABS+ have $D^{\text{sul/sil liq}} < 1000$ the TABS+ pattern at R=1 000 and R=10
545 000 is the same for As through to Bi. The difference is only evident from Cu onwards. As in
546 the case of the Kharaelakh and Talnakh ores the Noril'sk I ores also show Se and Te negative
547 anomalies, which we attribute to the magma having degassed.

548 **Conclusions**

549 We investigated the distribution of TABS+ in picrites and basalts from LIPs, komatiites and
550 MORB. Our main findings are summarized as follows:

- 551 1- Primitive mantle normalized plots can be used to consider the processes that affect
552 the distribution of TABS+.
- 553 2- Komatiites contain Se and Te at the same primitive mantle level as Pd, Pt and Cu,
554 and slightly higher than in the primitive mantle level of the highly incompatible
555 elements Th and Nb. This is consistent with derivation from a slightly depleted
556 mantle source with no sulfide phase retention during partial melting.

- 557 3- N-MORB contain As, Sb, Bi, Cu and Se at the same level of enrichment relative to
558 primitive mantle as Th, consistent with a depleted mantle source.
- 559 4- N-MORB are strongly depleted in Te, Pd and Pt, but not the other TABS and Cu.
560 This is consistent with segregation of a small amount of sulfide liquid.
- 561 5- E-MORB is similar to N-MORB with respect to the TABS+ and PGE, but enriched
562 in Th and Nb requiring a source that is not enriched in TABS, but is enriched in Th
563 and Nb.
- 564 6- The LIPs picrites are enriched in Th and Nb relative to the TABS+, Cu, Pd and Pt.
565 This enrichment combined with the high Mg# requires a peridotite source that has
566 been preferentially enriched with a small amount of a fluid or magma very rich in Th
567 **and** Nb, possibly a carbonatite.
- 568 7- Some LIPs basalts are enriched in Th over Nb due to assimilation of continental crust.
569 This would have also enriched the magmas in As, Sb and Bi. However, degassing
570 has led to the loss of As, Bi, Se and Te. Antimony does not appear to have been
571 significantly lost.
- 572 8- The initial concentrations of TABS+ in the magmas (komatiites and LIPs) from
573 which major Ni-Cu-PGE deposits form can be estimated as approximately twice
574 primitive mantle for komatiites and between 10 for As and Sb though 7 to 2 times
575 primitive mantle from Bi to Te for LIPs picrites.
- 576 9- The TABS+ and PGE contents of normal and thin Merensky reef and JM reef can be
577 modelled by collection of both TABS+ and PGE by a magmatic sulfide liquid in
578 equilibrium with a komatiitic magma contaminated with continental crust and do not
579 require TABS to collect the PGE.
- 580 10- The TABS+ concentrations in Noril'sk- Talnakh Ni-Cu deposits can be modelled as
581 being in equilibrium with a mafic magma that had been contaminated with

582 continental crust and which was also depleted in TABS+. This implies that the
583 magma was degassed or degassing at the time that the sulfides formed.

584

585 **Acknowledgements**

586 *This work was supported by a Canada Research Chair program grant to Sarah-Jane Barnes*
587 *(215503) and Discovery Grant 1884-2013. This manuscript benefited from insightful*
588 *comments from the reviewers, Drs. Giada Iacono-Marziano and Steve Barnes and careful*
589 *editorial handling by the editors.*

590

591

592 **References**

- 593 Anenburg, M., and Mavrogenes, J. A., 2016, Experimental observations on noble metal nanonuggets
594 and Fe-Ti oxides, and the transport of platinum group elements in silicate melts: *Geochimica*
595 *et Cosmochimica Acta*, v. 192, p. 258-278.
- 596 Arguin, J.-P., Pagé, P., Barnes, S.-J., Yu, S.-Y., and Song, X.-Y., 2016, The effect of chromite
597 crystallization on the distribution of osmium, iridium, ruthenium and rhodium in picritic
598 magmas: an example from the Emeishan Large Igneous Province, Southwestern China:
599 *Journal of Petrology*, v. 57, p. 1019-1048.
- 600 Barnes, S.-J., 2016, Chalcophile Elements, *in* White, W. M., ed., *Encyclopedia of Geochemistry: A*
601 *Comprehensive Reference Source on the Chemistry of the Earth*: Cham, Springer
602 International Publishing, p. 229-233.
- 603 Barnes, S.-J., and Lightfoot, P. C., 2005, Formation of magmatic nickel sulfide ore deposits and
604 processes affecting their copper and platinum group element contents, *in* Hedenquist, J. W.,
605 Thompson, J. F. H., Goldfarb, R. J., and Richards, J. P., eds., *One Hundredth Anniversary*
606 *Volume - Economic Geology 1905-2005*, Society of Economic Geologists, p. 179-213.
- 607 Barnes, S.-J., and Picard, C., 1993, The behaviour of platinum-group elements during partial melting,
608 crystal fractionation, and sulphide segregation: an example from the Cape Smith Fold Belt,
609 northern Quebec: *Geochimica et Cosmochimica Acta*, v. 57, p. 79-87.

610 Barnes, S.-J., and Ripley, E. M., 2016, Highly siderophile and strongly chalcophile elements in
611 magmatic ore deposits: *Reviews in Mineralogy and Geochemistry*, v. 81, p. 725-774.

612 Barnes, S.-J., Gorton, M., and Naldrett, A., 1983, A comparative study of olivine and clinopyroxene
613 spinifex flows from Alexo, Abitibi greenstone belt, Ontario, Canada: *Contributions to*
614 *Mineralogy and Petrology*, v. 83, p. 293-308.

615 Barnes, S.-J., Pagé, P., and Zientek, M., 2020, The Lower Banded series of the Stillwater Complex,
616 Montana: whole-rock lithophile, chalcophile, and platinum-group element distributions:
617 *Mineralium Deposita*, v. 55, p. 163-186.

618 Barnes, S. J., Cruden, A. R., Arndt, N., and Saumur, B. M., 2016, The mineral system approach
619 applied to magmatic Ni–Cu–PGE sulphide deposits: *Ore Geology Reviews*, v. 76, p. 296-
620 316.

621 Blanks, D. E., Holwell, D. A., Fiorentini, M. L., Moroni, M., Giuliani, A., Tassara, S., González-
622 Jiménez, J. M., Boyce, A. J., and Ferrari, E., 2020, Fluxing of mantle carbon as a physical
623 agent for metallogenic fertilization of the crust: *Nature communications*, v. 11, p. 1-11.

624 Boudreau, A., 1992, Volatile fluid overpressure in layered intrusions and the formation of
625 potholes: *Australian Journal of Earth Sciences*, v. 39, p. 277-287.

626 Boudreau, A. E., 2016, The Stillwater Complex, Montana—Overview and the significance of
627 volatiles: *Mineralogical Magazine*, v. 80, p. 585-637.

628 Brenan, J. M., 2015, Se–Te fractionation by sulfide–silicate melt partitioning: Implications for the
629 composition of mantle-derived magmas and their melting residues: *Earth and Planetary*
630 *Science Letters*, v. 422, p. 45-57.

631 Brüggemann, G., Naldrett, A., Asif, M., Lightfoot, P., Gorbachev, N., and Fedorenko, V., 1993,
632 Siderophile and chalcophile metals as tracers of the evolution of the Siberian Trap in the
633 Noril'sk region, Russia: *Geochimica et Cosmochimica Acta*, v. 57, p. 2001-2018.

634 Cafagna, F., and Jugo, P. J., 2016, An experimental study on the geochemical behavior of highly
635 siderophile elements (HSE) and metalloids (As, Se, Sb, Te, Bi) in a mss-iss-pyrite system at
636 650° C: a possible magmatic origin for Co-HSE-bearing pyrite and the role of metalloid-rich
637 phases in the fractionation of HSE: *Geochimica et Cosmochimica Acta*, v. 178, p. 233-258.

638 Campbell, I. H., and Barnes, S. J., 1984, A model for the geochemistry of the platinum-group
639 elements in magmatic sulphide deposits: *Canadian Mineralogist*, v. 22, p. 151-160.

640 Canali, A.C., Brenan, J.M., and Sullivan, N.A., 2017, Solubility of platinum-arsenide melt and
641 sperrylite in synthetic basalt at 0.1 MPa and 1200° C with implications for arsenic speciation
642 and platinum sequestration in mafic igneous systems: *Geochimica et Cosmochimica Acta*, v.
643 216, p. 153-168.

644 Cox, D., Watt, S. F., Jenner, F. E., Hastie, A. R., and Hammond, S. J., 2019, Chalcophile element
645 processing beneath a continental arc stratovolcano: *Earth and Planetary Science Letters*, v.
646 522, p. 1-11.

647 Dare, S. A., Barnes, S.-J., Prichard, H. M., and Fisher, P. C., 2014, Mineralogy and geochemistry of
648 Cu-rich ores from the McCreedy East Ni-Cu-PGE deposit (Sudbury, Canada): implications
649 for the behavior of platinum group and chalcophile elements at the end of crystallization of
650 a sulfide liquid: *Economic Geology*, v. 109, p. 343-366.

651 Distler, V. V., 1994, Platinum Mineralisation of the Noril'sk Deposits, *in* Lightfoot, P. C., and
652 Naldrett, A. J., eds., *Proceedings of the Sudbury-Noril'sk symposium*, Special Publication
653 no. 5, Ontario Geological Survey, p. 243-262.

654 Duran, C. J., Barnes, S.-J., Pleše, P., Prašek, M. K., Zientek, M. L., and Pagé, P., 2017, Fractional
655 crystallization-induced variations in sulfides from the Noril'sk-Talnakh mining district
656 (polar Siberia, Russia): *Ore Geology Reviews*, v. 90, p. 326-351.

657 Edmonds, M., and Mather, T. A., 2017, Volcanic sulfides and outgassing: *Elements*, v. 13, p. 105-
658 110.

659 Edmonds, M., Mather, T. A., and Liu, E. J., 2018, A distinct metal fingerprint in arc volcanic
660 emissions: *Nature Geoscience*, v. 11, p. 790-794.

661 Forrest, A., Keller, K., and Schilling, J.-G., 2017, Selenium, tellurium and sulfur variations in basalts along the
662 Reykjanes Ridge and extension over Iceland, from 50 N to 65 N: *Interdisciplinary Earth Data Alliance*
663 (IEDA), Palisades, NY, doi, v. 10.

664 Godel, B., González-Álvarez, I., Barnes, S. J., Barnes, S.-J., Parker, P., and Day, J., 2012, Sulfides
665 and sulfarsenides from the rosie nickel prospect, Duketon Greenstone Belt, Western
666 Australia: *Economic Geology*, v. 107, p. 275-294.

667 Greaney, A. T., Rudnick, R. L., Helz, R. T., Gaschnig, R. M., Piccoli, P. M., and Ash, R. D., 2017,
668 The behavior of chalcophile elements during magmatic differentiation as observed in Kilauea
669 Iki lava lake, Hawaii: *Geochimica et Cosmochimica Acta*, v. 210, p. 71-96.

670 Guo, H., and Audétat, A., 2017, Transfer of volatiles and metals from mafic to felsic magmas in
671 composite magma chambers: an experimental study: *Geochimica et Cosmochimica Acta*, v.
672 198, p. 360-378.

673 Hammerli, J., Spandler, C., and Oliver, N. H., 2016, Element redistribution and mobility during upper
674 crustal metamorphism of metasedimentary rocks: an example from the eastern Mount Lofty
675 Ranges, South Australia: *Contributions to Mineralogy and Petrology*, v. 171, p. 36.

676 Hanley, J. J., 2007, The role of arsenic-rich melts and mineral phases in the development of high-
677 grade Pt-Pd mineralization within komatiite-associated magmatic Ni-Cu sulfide horizons at
678 Dundonald Beach South, Abitibi subprovince, Ontario, Canada: *Economic Geology*, v.
679 102(2), p. 305-317.

680 Hattori, K. H., Arai, S., and Clarke, D. B., 2002, Selenium, tellurium, arsenic and antimony contents
681 of primary mantle sulfides: *The Canadian Mineralogist*, v. 40, p. 637-650.

682 Helmy, H. M., Ballhaus, C., Berndt, J., Bockrath, C., and Wohlgemuth-Ueberwasser, C., 2007,
683 Formation of Pt, Pd and Ni tellurides: experiments in sulfide–telluride systems:
684 Contributions to Mineralogy and Petrology, v. 153, p. 577-591.

685 Helmy, H. M., Ballhaus, C., Wohlgemuth-Ueberwasser, C., Fonseca, R. O., and Laurenz, V., 2010,
686 Partitioning of Se, As, Sb, Te and Bi between monosulfide solid solution and sulfide melt–
687 application to magmatic sulfide deposits: Geochimica et Cosmochimica Acta, v. 74, p. 6174-
688 6179.

689 Helmy, H. M., Ballhaus, C., Fonseca, R. O., and Leitzke, F. P., 2020, Concentrations of Pt, Pd, S,
690 As, Se and Te in silicate melts at sulfide, arsenide, selenide and telluride saturation: evidence
691 of PGE complexing in silicate melts?: Contributions to Mineralogy and Petrology, v. 175, p.
692 1-14..

693 Holwell, D. A., Fiorentini, M., McDonald, I., Lu, Y., Giuliani, A., Smith, D. J., Keith, M., and
694 Locmelis, M., 2019, A metasomatized lithospheric mantle control on the metallogenic
695 signature of post-subduction magmatism: Nature communications, v. 10, p. 1-10.

696 Hu, Z., and Gao, S., 2008, Upper crustal abundances of trace elements: a revision and update:
697 Chemical Geology, v. 253, p. 205-221.

698 Iacono-Marziano, G., Marecal, V., Pirre, M., Gaillard, F., Arteta, J., Scaillet, B., and Arndt, N. T.,
699 2012, Gas emissions due to magma–sediment interactions during flood magmatism at the
700 Siberian Traps: Gas dispersion and environmental consequences: Earth and Planetary
701 Science Letters, v. 357, p. 308-318.

702 Iacono-Marziano, G., Ferraina, C., Gaillard, F., Di Carlo, I., Arndt, N. T., 2017, Assimilation of
703 sulfate and carbonaceous rocks: Experimental study, thermodynamic modeling and
704 application to the Noril’sk-Talnakh region (Russia): Ore Geology Reviews, v. 90, p. 399-
705 413.

706 Jenkins, M. C., and Mungall, J. E., 2018, Genesis of the peridotite zone, Stillwater Complex,
707 Montana, USA: Journal of Petrology, v. 59, p. 2157-2189.

708 Jenkins, M.C., Mungall, J. E., Zientek, M. L., Holick, P., and Butak, K., 2020, The Nature and
709 Composition of the JM Reef, Stillwater Complex, Montana, USA: Economic Geology, v.
710 115, p. 1799-1826.

711 Jenner, F. E., and O'Neill, H. S. C., 2012, Analysis of 60 elements in 616 ocean floor basaltic glasses:
712 Geochemistry, Geophysics, Geosystems, v. 13, Q0200.

713 Kamenetsky, V. S., and Eggins, S. M., 2012, Systematics of metals, metalloids, and volatiles in
714 MORB melts: effects of partial melting, crystal fractionation and degassing (a case study of
715 Macquarie Island glasses): Chemical Geology, v. 302, p. 76-86.

716 Kamenetsky, V. S., Chung, S.-L., Kamenetsky, M. B., and Kuzmin, D. V., 2012, Picrites from the
717 Emeishan Large Igneous Province, SW China: a compositional continuum in primitive
718 magmas and their respective mantle sources: Journal of Petrology, v. 53, p. 2095-2113.

- 719 Ketris, M., and Yudovich, Y. E., 2009, Estimations of Clarkes for Carbonaceous biolithes: World
720 averages for trace element contents in black shales and coals: *International Journal of Coal*
721 *Geology*, v. 78, p. 135-148.
- 722 Kinloch, E. D., 1982, Regional trends in the platinum-group mineralogy of the critical zone of the
723 Bushveld Complex, South Africa: *Economic Geology*, v. 77(6), p. 1328-1347.
- 724 Le Vaillant, M., Barnes, S. J., Fiorentini, M. L., Barnes, S.-J., Bath, A., and Miller, J., 2018,
725 Platinum-group element and gold contents of arsenide and sulfarsenide minerals associated
726 with Ni and Au deposits in Archean greenstone belts: *Mineralogical Magazine*, v. 82, p. 625-
727 647.
- 728 Li, Y., and Audétat, A., 2015, Effects of temperature, silicate melt composition, and oxygen fugacity
729 on the partitioning of V, Mn, Co, Ni, Cu, Zn, As, Mo, Ag, Sn, Sb, W, Au, Pb, and Bi between
730 sulfide phases and silicate melt: *Geochimica Et Cosmochimica Acta*, v. 162, p. 25-45.
- 731 Liang, Q.-L., Song, X.-Y., Wirth, R., Chen, L.-M., and Dai, Z.-H., 2019, Implications of nano-and
732 micrometer-size platinum-group element minerals in base metal sulfides of the Yangliuping
733 Ni-Cu-PGE sulfide deposit, SW China: *Chemical Geology*, v. 517, p. 7-21.
- 734 Lightfoot, P. C., and Keays, R. R., 2005, Siderophile and chalcophile metal variations in flood basalts
735 from the Siberian trap, Noril'sk region: Implications for the origin of the Ni-Cu-PGE sulfide
736 ores: *Economic Geology*, v. 100, p. 439-462.
- 737 Lightfoot, P., Hawkesworth, C., Hergt, J., Naldrett, A., Gorbachev, N., Fedorenko, V., and Doherty,
738 W., 1993, Remobilisation of the continental lithosphere by a mantle plume: major-, trace-
739 element, and Sr-, Nd-, and Pb-isotope evidence from picritic and tholeiitic lavas of the
740 Noril'sk District, Siberian Trap, Russia: *Contributions to Mineralogy and Petrology*, v. 114,
741 p. 171-188.
- 742 Liu, Y., and Brenan, J., 2015, Partitioning of platinum-group elements (PGE) and chalcogens (Se,
743 Te, As, Sb, Bi) between monosulfide-solid solution (MSS), intermediate solid solution (ISS)
744 and sulfide liquid at controlled fO_2 - fS_2 conditions: *Geochimica et Cosmochimica Acta*, v.
745 159, p. 139-161.
- 746 Lodders, K., 2003, Solar system abundances and condensation temperatures of the elements:
747 *Astrophysical Journal*, v. 591, p. 1220-1247.
- 748 Lyubetskaya, T., and Korenaga, J., 2007, Chemical composition of Earth's primitive mantle and its
749 variance: 1. Method and results: *Journal of Geophysical Research: Solid Earth (1978–2012)*,
750 v. 112., B03211
- 751 Maciag, B. J., and Brenan, J. M., 2020, Speciation of arsenic and antimony in basaltic magmas:
752 *Geochimica et Cosmochimica Acta*, v. 276, p. 198-218.
- 753 Maier, W. D., Barnes, S.-J., and Marsh, J. S., 2003, The concentrations of the noble metals in
754 Southern African flood-type basalts and MORB: implications for petrogenesis and magmatic
755 sulphide exploration: *Contributions to Mineralogy and Petrology*, v. 146, p. 44-61

756 Maier, W. D., Barnes, S. J., Campbell, I. H., Fiorentini, M. L., Peltonen, P., Barnes, S.-J., and
757 Smithies, R. H., 2009, Progressive mixing of meteoritic veneer into the early Earth's deep
758 mantle: *Nature*, v. 460, p. 620-623.

759 Maier, W., Barnes, S.-J., and Groves, D., 2013, The Bushveld Complex, South Africa: formation of
760 platinum–palladium, chrome-and vanadium-rich layers via hydrodynamic sorting of a
761 mobilized cumulate slurry in a large, relatively slowly cooling, subsiding magma chamber:
762 *Mineralium Deposita*, v. 48, p. 1-56.

763 Mansur, E. T., and Barnes, S.-J., 2020a, The role of Te, As, Bi, Sn and Sb during the formation of
764 platinum-group-element reef deposits: Examples from the Bushveld and Stillwater
765 Complexes: *Geochimica et Cosmochimica Acta*, v. 272, p. 235-258.

766 Mansur, E. T., and Barnes, S.-J., 2020b, Concentrations of Te, As, Bi, Sb and Se in the Marginal
767 Zone of the Bushveld Complex: Evidence for crustal contamination and the nature of the
768 magma that formed the Merensky Reef: *Lithos*, v. 358-359,

769 Mansur, E. T., Barnes, S.-J., Duran, C. J., and Sluzhenikin, S. F., 2020a, Distribution of chalcophile
770 and platinum-group elements among pyrrhotite, pentlandite, chalcopyrite and cubanite from
771 the Noril'sk-Talnakh ores: Implications for the formation of platinum-group minerals:
772 *Mineralium Deposita*, v. 55, p. 1215-1232.

773 Mansur, E. T., Barnes, S. J., Savard, D., and Webb, P. C., 2020b, Determination of Te, As, Bi, Sb
774 and Se (TABS) in Geological Reference Materials and GeoPT Proficiency Test Materials by
775 Hydride Generation-Atomic Fluorescence Spectrometry (HG-AFS): *Geostandards and*
776 *Geoanalytical Research*, v. 44, p. 147-167.

777 Marsh, J., Hooper, P., Rehacek, J., Duncan, R., and Duncan, A., 1997, Stratigraphy and age of Karoo
778 basalts of Lesotho and implications for correlations within the Karoo igneous province:
779 *Geophysical Monograph-American Geophysical Union*, v. 100, p. 247-272

780 Mather, T. A., 2015, Volcanoes and the environment: Lessons for understanding Earth's past and
781 future from studies of present-day volcanic emissions: *Journal of Volcanology and*
782 *Geothermal Research*, v. 304, p. 160-179.

783 McDonough, W. F., and Arevalo Jr, R., 2008, Uncertainties in the composition of Earth, its core and
784 silicate sphere: *Journal of Physics: Conference Series*, 2008, p. 022006.

785 Naldrett, A. J., 2011, Fundamentals of magmatic sulfide deposit, *in* Li, C., and Ripley, E. M., eds.,
786 *Magmatic Ni-Cu and PGE deposits: Geology, Geochemistry, and Genesis: Society of*
787 *Economic Geologists*, p. 1-50.

788 Palme, H., and O'Neill, H. St. C., 2014, Cosmochemical estimates of mantle composition., *in*
789 Carlson, R. W., ed., *Treatise on Geochemistry*, Vol. 3., Elsevier, p. 1–39

790 Patten, C., Barnes, S.-J., Mathez, E.A., and Jenner, F.E., 2013, Partition coefficients of chalcophile
791 elements between sulfide and silicate melts and the early crystallization history of sulfide

792 liquid: LA-ICP-MS analysis of MORB sulfide droplets: *Chemical Geology*, v. 358, p. 170-
793 188.

794 Patten, C.G., Pitcairn, I.K., and Teagle, D.A.H., 2017, Hydrothermal mobilisation of Au and other
795 metals in supra-subduction oceanic crust: Insights from the Troodos ophiolite: *Ore Geology*
796 *Reviews*, v. 86, p. 487-508.

797 Picard, C., Lamothe, D., Piboule, M., and Oliver, R., 1990, Magmatic and geotectonic evolution of
798 a Proterozoic oceanic basin system: The Cape Smith thrust-fold belt (New-Quebec):
799 *Precambrian Research*, v. 47(3-4), p. 223-249.

800 Piña, R., Gervilla, F., Barnes, S-J., Ortega, L., and Lunar, R., 2013, Partition coefficients of platinum
801 group and chalcophile elements between arsenide and sulfide phases as determined in the
802 Beni Bousera Cr-Ni mineralization (North Morocco): *Economic Geology*, v. 108(5), p. 935-
803 951.

804 Piña, R., Gervilla, F., Barnes, S-J., Ortega, L., and Lunar, R., 2015, Liquid immiscibility between
805 arsenide and sulfide melts: evidence from a LA-ICP-MS study in magmatic deposits at
806 Serranía de Ronda (Spain): *Mineralium Deposita*, v. 50(3), p. 265-279.

807 Pitcairn, I. K., Craw, D., and Teagle, D. A., 2015, Metabasalts as sources of metals in orogenic gold deposits:
808 *Mineralium Deposita*, v. 50, p. 373-390.

809 Pokrovski, G. S., Borisova, A. Y., and Bychkov, A. Y., 2013, Speciation and transport of metals and metalloids in
810 geological vapors: *Reviews in Mineralogy and Geochemistry*, v. 76, p. 165-218.

811 Roberts, M., Reid, D., Miller, J., Basson, I., Roberts, M., and Smith, D., 2007, The Merensky Cyclic
812 Unit and its impact on footwall cumulates below Normal and Regional Pothole reef types in
813 the Western Bushveld Complex: *Mineralium Deposita*, v. 42, p. 271-292.

814 Samalens, N., Barnes, S-J., and Sawyer, E.W., 2017, The role of black shales as a source of sulfur
815 and semimetals in magmatic nickel-copper deposits: Example from the Partridge River
816 Intrusion, Duluth Complex, Minnesota, USA: *Ore Geology Reviews*, v. 81(1), p. 173-187.

817 Shellnutt, J.G., 2013, The Emeishan large igneous province: a synthesis: *Geoscience Frontiers*, v. 5,
818 p. 369-394.

819 Shevko, E.P., Bortnikova, S.B., Abrosimova, N.A., Kamenetsky, V.S., Bortnikova, S.P., Panin, G.L.,
820 and Zelenski M., 2018, Trace elements and minerals in fumarolic sulfur: the case of Ebeko
821 volcano, Kurile: *Geofluids* p. 16.

822 Sinyakova, E., Kosyakov, V., Borisenko, A., 2017, Effect of the presence of As, Bi, and Te on the
823 behavior of Pt metals during fractionation crystallization of sulfide magma: *Doklady Earth*
824 *Sciences*, v. 477, p.1422–1425.

825 Sluzhenikin, S. F., Krivolutskaya, N. A., Rad'ko, V. A., Malitch, K. N., Distler, V. A., and
826 Fedorenko, V. A., 2014, Ultramafic-mafic intrusions, volcanic rocks and PGE-Cu-Ni sulfide
827 deposits of the Noril'sk Province, Polar Siberia, *in* Field trip guidebook, 12th International
828 Platinum Symposium Yekaterinburg, p. 80.

- 829 Steele, T. W., Levin, J., and Copelwitz, I., 1975, Preparation and certification of a reference sample
830 of a precious metal ore: National Institute for Metallurgy, p. 1696-1975.
- 831 Stucker, V. K., Walker, S. L., de Ronde, C. E., Caratori Tontini, F., and Tsuchida, S., 2017,
832 Hydrothermal Venting at Hinepuia Submarine Volcano, Kermadec Arc: Understanding
833 Magmatic-Hydrothermal Fluid Chemistry: Geochemistry, Geophysics, Geosystems, v.
834 18(10), p. 3646-3661.
- 835 Viljoen, M., 1999, The nature and origin of the Merensky Reef of the western Bushveld Complex
836 based on geological facies and geophysical data: South African Journal of Geology, v. 102,
837 p. 221-239.
- 838 Wang, H. S., Lineweaver, C. H., and Ireland, T. R., 2018, The elemental abundances (with
839 uncertainties) of the most Earth-like planet: Icarus, v. 299, p. 460-474.
- 840 Wieser, P., Jenner, F., Edmonds, M., MacLennan, J., and Kunz, B., 2020, Chalcophile elements track
841 the fate of sulfur at Kīlauea Volcano, Hawai'i: Geochimica et Cosmochimica Acta, v. 282,
842 p. 245-275.
- 843 Wilson, A. H., 2012, A Chill Sequence to the Bushveld Complex: Insight into the First Stage of
844 Emplacement and Implications for the Parental Magmas: Journal of Petrology, v. 53, p.
845 1123-1168.
- 846 Wood, B. J., Smythe, D. J., and Harrison, T., 2019, The condensation temperatures of the elements:
847 a reappraisal: American Mineralogist: Journal of Earth and Planetary Materials, v. 104, p.
848 844-856.
- 849 Yi, W., Halliday, A.N., Alt, J.C., Lee, D.C., Rehkämper, M., Garcia, M.O., and Su, Y., 2000,
850 Cadmium, indium, tin, tellurium, and sulfur in oceanic basalts: Implications for chalcophile
851 element fractionation in the Earth: Journal of Geophysical Research: Solid Earth, v. 105(B8),
852 p. 18927-18948.
- 853 Zelenski, M., Malik, N., and Taran, Y., 2014, Emissions of trace elements during the 2012–2013
854 effusive eruption of Tolbachik volcano, Kamchatka-enrichment factors, partition
855 coefficients and aerosol contribution. Journal of Volcanology and Geothermal Research, v.
856 285, p. 136-149.
- 857 Zhang, L., Ren, Z. Y., Handler, M. R., Wu, Y. D., Zhang, L., Qian, S. P., and Xu, Y. G., 2019, The
858 origins of high-Ti and low-Ti magmas in large igneous provinces, insights from melt
859 inclusion trace elements and Sr-Pb isotopes in the Emeishan large Igneous
860 Province: Lithos, v. 344, p. 122-133.
- 861 Zientek, M. L., 2012, Magmatic ore deposits in layered intrusions - Descriptive model for reef-type
862 PGE and contact-type Cu-Ni-PGE deposits: U.S. Geological Survey Open File, 2012-1010,
863 p. 48.

864 **Figure captions**

865 **Figure 1** – Location of samples used in this study.

866 **Figure 2** – A) Primitive mantle (Lyubetskaya and Korenaga, 2007) normalized plots illustrating the
867 effects of; sulfide segregation, sulfide accumulation, degassing, partial melting and crystal
868 fractionation. B) Primitive mantle normalized plot of the Bushveld B-1 magma (Mansur and Barnes,
869 2020b) illustrating the effects of assimilation of continental crustal. Note the negative Nb anomaly
870 and enrichment of As, Sb and Bi relative to the other chalcophile elements in black shale (Samalens
871 et al. 2017; Ketris and Yudovich, 2009) and upper continental crust (Hu and Gao, 2008) which are
872 also present in the B-1 magma.

873 **Figure 3** – Median primitive mantle normalized plots of: A) MORB samples, note that all samples
874 are depleted in Te, Pd and Pt and that relative to N-MORB E-MORB shows enrichment in both Th
875 and Nb, but not in the chalcophile elements; B) Basalts from the Cape Smith fold belt, note the flat
876 pattern; C) Komatiites from the Abitibi and Baby-Belleterre Greenstone Belts; D) Komatiites from
877 the Barberton Greenstone Belt; E) Picrites and basalts from Emeishan Province; F) Picrites and
878 dolerites from Etendeka Province; G) Picrite and basalts from Karoo Province; H) Basalts from the
879 Siberian Province. Note for all of the rocks from large igneous provinces have negative As, Bi, Te
880 and Se anomalies thought to be the product of degassing.

881 **Figure 4.** Mg# versus A) Nb and B) Bi for the Alexo komatiite flows. Note Nb and Mg# show a
882 negative correlation with the olivine rich parts of the flow being depleted in Nb as would be expected
883 during crystal fractionation. In contrast, Bi does not show a coherent pattern.

884 **Figure 5** – Primitive mantle normalized plots comparing models of crustal assimilation with the
885 observed patterns: A) Picrite dike and dolerite from the Etendeka Province and B) Picrite and basalts
886 from Tuli (Karoo Province).

887 **Figure 6** – Comparison of primitive mantle normalized patterns of all the picrites. Note also none of
888 the picrites have negative Nb anomalies indicating that the component that enriched the mantle in
889 incompatible lithophile elements was not sedimentary. Furthermore, the level and shape of the
890 patterns for all of the picrites are similar for the chalcophile elements and the level of the patterns is

891 lower than that of Th and Nb indicating that the component enriching the source in lithophile
892 elements did not enrich the source in chalcophile elements.

893 **Figure 7.** Primitive mantle normalized plots of: A) Normal and thin Merensky reef (MR; Bushveld
894 Complex); B) Composite Merensky and Plat reefs (Bushveld Complex); C) JM reef (Stillwater
895 Complex). Data from Mansur and Barnes (2020a) and African Minerals Standards www.amis.co.za/

896 **Figure 8.** Primitive mantle normalized plots of: A) Cu-poor B) Cu-rich and C) Transitional ore from
897 Talnakh and Kharaelakh intrusions and D) Cu-rich and transitional ore from the Noril'sk 1 intrusion.
898 Data from Duran et al. (2017).

899 **Table caption**

900 **Table 1** – Median Values of Chalcophile Elements, Th, Nb and Mg# for studied rocks.

901 **Table 2** – Estimations of TABS concentrations in komatiites and in large igneous provinces
902 picrites.

903

904 *Online resources – Data availability*

905 **Appendices**

906 **Appendix 1** Sample descriptions and analytical methods

907 **Appendix 2 Table A1** – Analyses of reference materials used to monitor the data quality of HG-
908 AFS analyses. **Table A2** – Analyses of reference materials used to monitor the data quality of
909 whole-rock analyses. **Table A3** – Whole-rock results obtained in this study and compiled from
910 previous studies.

911 **Appendix 3 Figure A1** – Primitive mantle normalized Th, Nb, Cu, TABS, Pd and Pt plots: **A** high-
912 Ti picrites – Emeishan Province; **B** low-Ti picrites – Emeishan Province; **C** Subvolcanic sill –
913 Emeishan Province; **D** basalts – Emeishan Province; **E** Lesotho Formation – Karoo Province; **F**
914 Barkly East Formation – Karoo Province; **G** Tuli Formation – Karoo Province; **H** Siberian Traps.

915 Primitive mantle values from Lyubetskaya and Korenaga (2007) *Figure A2* – Primitive mantle
916 normalized Th, Nb, Cu, TABS, Pd and Pt plots: **A** Alexo – Abitibi Greenstone belt **B** Cape Smith
917 belt; **C** Baby Formation; Primitive mantle values from Lyubetskaya and Korenaga (2007)

918

Table 1		Median Values of Chalcophile Elements, Th, Nb and Mg# for Rocks Presented in This Study													
Location or Province	Formation or Location	Rock type	n	As ppm	Se ppm	Sb ppm	Te ppm	Bi ppm	S ppm	Cu ppm	Pd ppb	Pt ppb	Nb ppm	Th ppm	Mg#
MORB															
South Atlantic Ridge		N MORB	6	0.157	0.222	0.019	0.008	0.007	1264	79	0.32	0.05	1.5	0.22	0.63
Hotu Sea Mount		E MORB	2	0.188	0.226	0.020	0.007	0.010	1229	85	0.79	0.35	22.5	0.9	0.58
Garret fracture zone		N MORB	3	0.161	0.225	0.019	0.009	0.010	1167	76	0.35	0.18	4.6	0.18	0.65
Cape Smith	cycle d + e	MORB like	4	0.327	0.301	0.068	0.026	0.053	250	110	12.42	14.07	3.9	0.33	0.61
Komatiites															
Abitibi	Alexo	Spinifex	8	0.405	0.164	0.201	0.020	0.053	450	46	10.09	10.25	0.5	0.06	0.84
Abitibi	Alexo	B2	6	0.266	0.137	0.190	0.014	0.019	500	34	6.77	8.36	0.4	0.05	0.86
Belletere	Baby	Pillows and massive	9	0.836	0.098	0.186	0.007	0.028	349	20	14.00	14.00	1.5	0.05	0.79
Barberton	Komati	Spinifex	2	0.668	0.056	0.147	0.007	0.023	147	31	7.06	5.97	1.9	0.26	0.80
Barberton	Sandspruit	Spinifex	2	1.260	0.024	0.142	<0.005	0.013	n.d.	5	2.72	2.44	1.4	0.15	0.83
Barberton	Hooggenog	Spinifex	2	0.707	0.034	0.127	<0.005	0.019	n.d.	104	4.73	4.72	4.3	1.15	0.75
Picrites															
Emeishan	Daju	High-Ti	6	0.308	0.023	0.073	<0.005	0.010	247	103	2.79	6.78	14.8	1.61	0.78
Emeishan	Shiman	High-Ti	2	0.340	0.024	0.071	<0.005	0.022	139	104	3.42	9.67	15.2	1.6	0.78
Emeishan	Jianchuan	High-Ti	7	0.263	0.038	0.084	0.008	0.007	64	106	7.04	8.35	16.8	1.86	0.76
Emeishan	Dali	Low-Ti	4	0.219	0.046	0.063	0.008	0.039	40	124	5.78	7.17	4.8	0.68	0.77
Emeishan	Binchuan	Low-Ti	5	0.636	0.014	0.094	<0.005	0.014	40	101	5.79	7.81	11.7	2.04	0.80
Etendeka	Horingbaai	Low-Ti	1	0.044	0.040	0.054	<0.005	0.023	150	99	7.00	6.00	4.9	0.65	0.77
Etendeka	Tafelberg	Low-Ti	2	0.198	0.107	0.092	0.019	0.034	505	88	12.00	8.00	4.5	n.a.	0.74
Karoo	Tuli	High-Ti	1	0.502	0.029	0.105	<0.005	<0.005	580	72	4.00	9.00	14.9	2.29	0.74
Basalts and Dolerites															
Emeishan	all 5 localities	Hi-Ti basalt	9	0.392	0.028	0.099	<0.005	<0.005	40	140	6.63	7.82	26.2	2.31	0.60
Etendeka	Horingbaai	Lo-Ti dolerite	2	0.270	0.121	0.076	0.017	0.050	265	239	15.00	6.00	14.0	1.74	0.53
Etendeka	Tafelberg	Lo-Ti dolerite	1	2.076	0.157	0.236	0.024	0.076	650	249	10.00	6.00	11.8	4.77	0.35
Karoo	Lesotho	Lo-Ti basalt	12	0.350	0.031	0.088	<0.005	<0.005	130	97	6.00	5.00	6.8	1.52	0.52
Karoo	Barkly East	Lo-Ti basalt	9	0.590	0.050	0.119	0.008	0.018	254	133	3.50	2.00	6.0	2.05	0.56
Karoo	Tuli	Hi-Ti basalt	2	0.948	0.037	0.151	0.010	0.025	40	281	13.00	8.00	18.0	3.53	0.44
Siberia	Nd	Lo-Ti basalt	2	0.617	0.041	0.185	<0.005	0.029	193	73	0.37	0.50	8.1	3.53	0.56
Siberia	Mr-Kh	Lo-Ti basalt	3	0.368	0.103	0.157	0.009	<0.005	110	238	9.99	10.05	5.1	1.32	0.56
Miscellaneous															
Emeishan	Ertan	Hi-Ti sill	3	0.418	0.035	0.095	<0.005	<0.005	88	87	7.95	18.81	25.9	3.55	0.77
Siberia	Medvezhy Ruchei open pit	Shale	1	10.922	0.031	0.397	0.008	0.127	730	154	4.27	1.67	22.7	19.29	0.52
SARM-7	Bushveld	Merensky reef composite	1	1.38	2.44	1.62	0.62	0.64	4170	700	1542	3700	1.0	1	0.77
ECBV109	Bushveld	Hornfels sediment	1	4.88	0.068	0.82	0.006	0.032	<40	14	nd	nd	12.4	16.2	0.58
n = number of samples	n.d = not determined														

919

920

Table 2	Estimations of TABS concentrations in komatiites and LIPS picrites.							
Element	As	Sb	Bi	Cu	Se	Te	Pd	Pt
units	ppm	ppm	ppm	ppm	ppm	ppm	ppb	ppb
Al-undepleted komatiites	0.1*	0.014*	0.008*	46+	0.16+	0.02+	10+	10+
Lo-Ti picrites	0.5*	0.08#	0.03#	103#	0.26#	0.01#	7.2#	7.6#
Hi-Ti picrite	0.5*	0.08#	0.03#	96#	0.26#	0.006#	4.3#	8.7#
* = estimate based on mantle normalization								
+ = average of Alexo spinifex for komatiites; # = average of picrites								

921

922

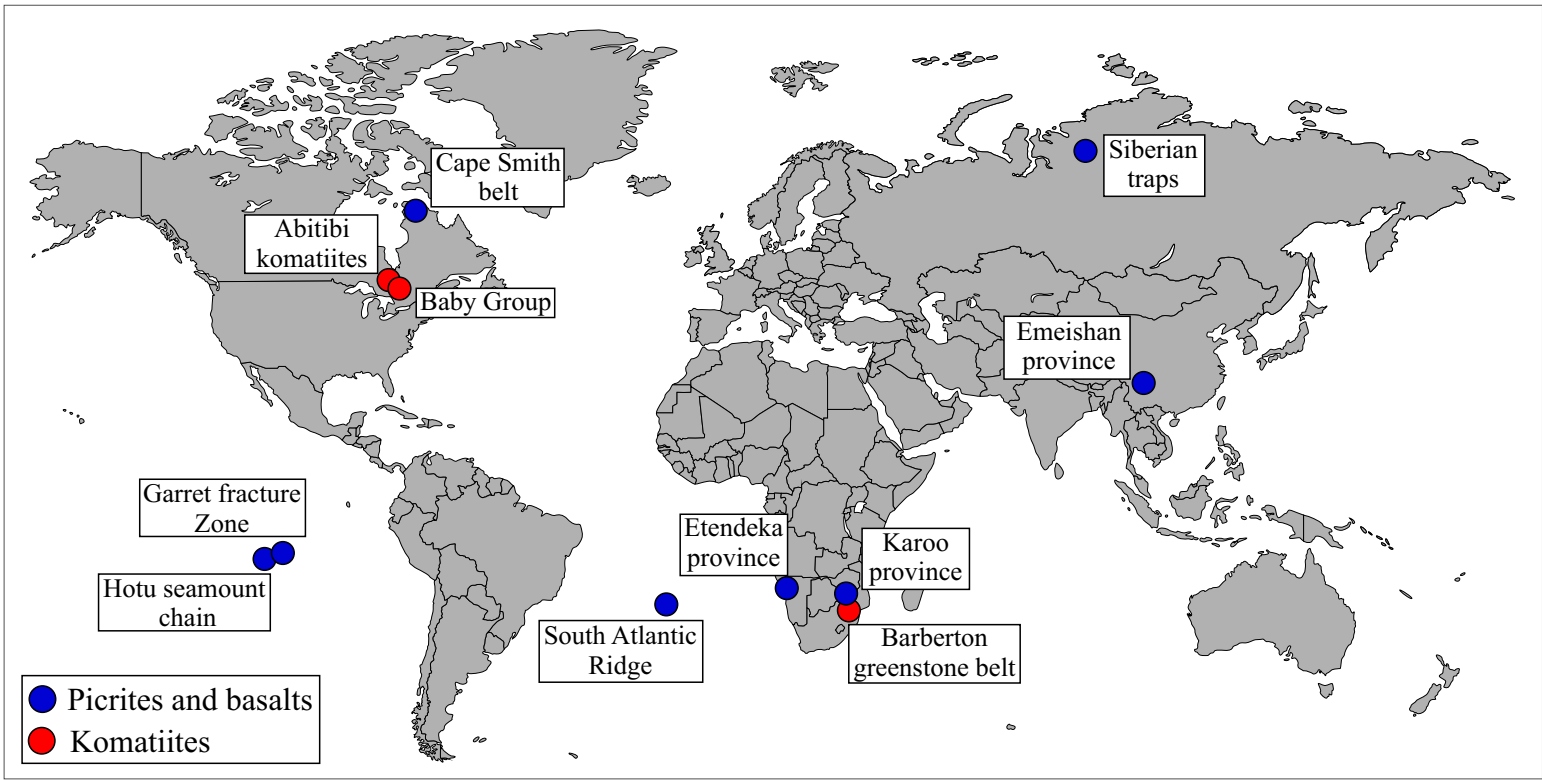


Fig.1

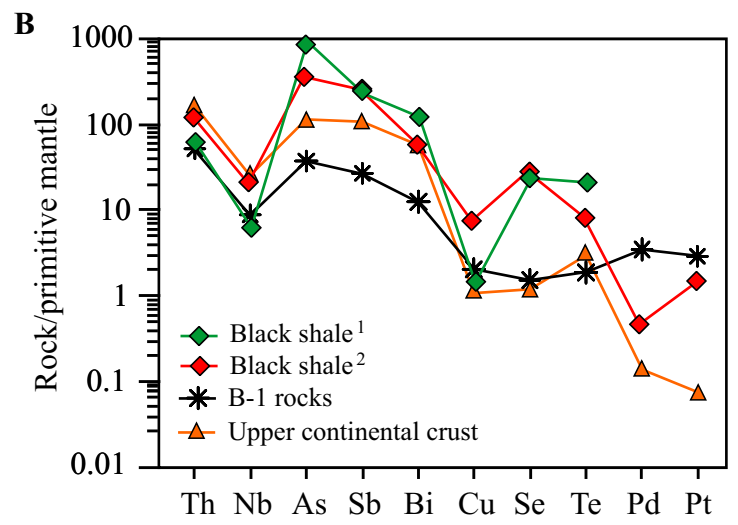
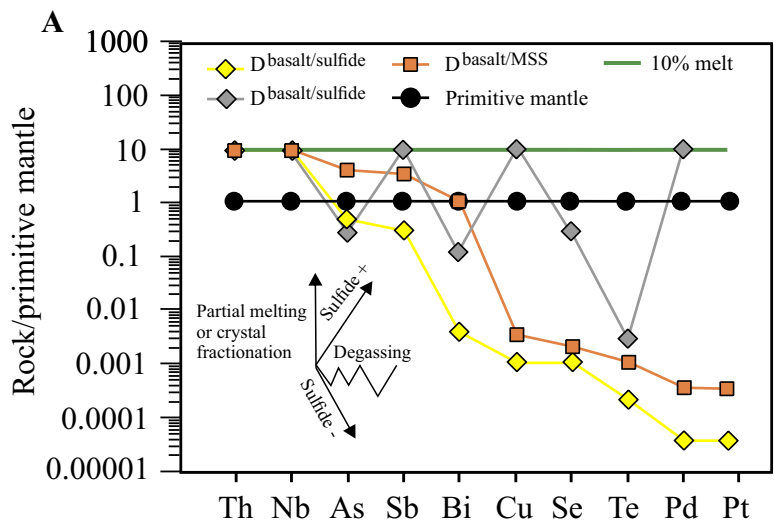
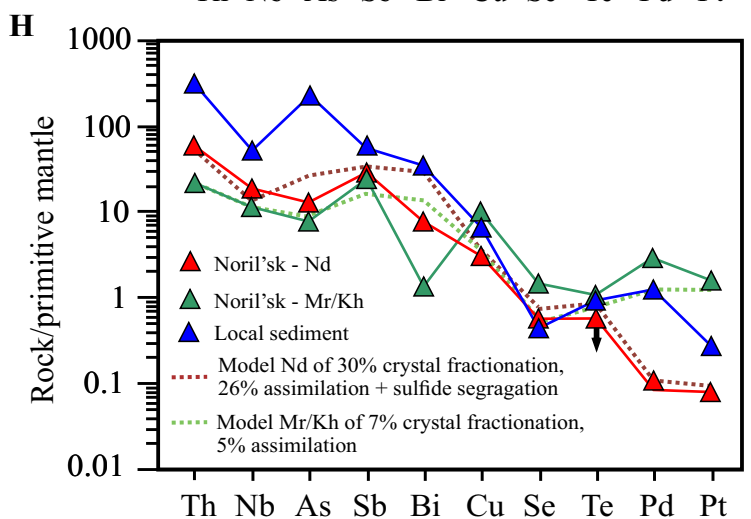
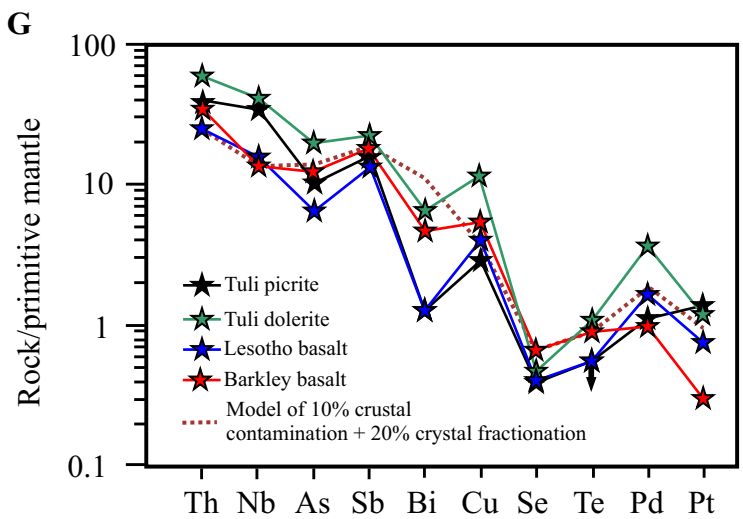
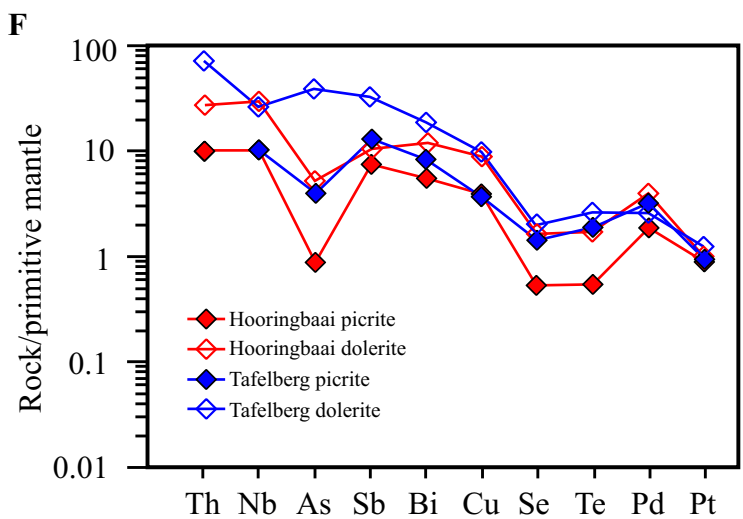
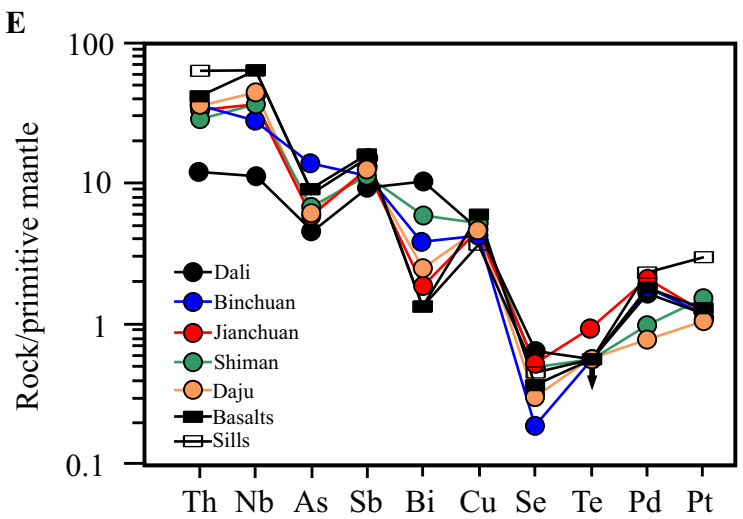
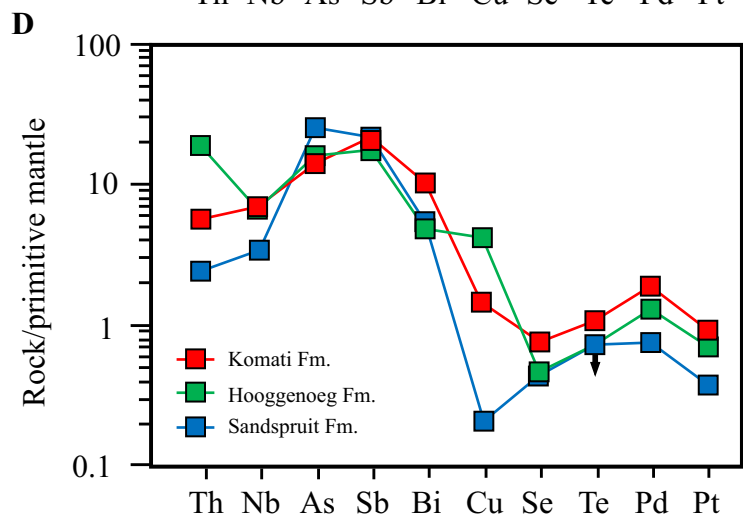
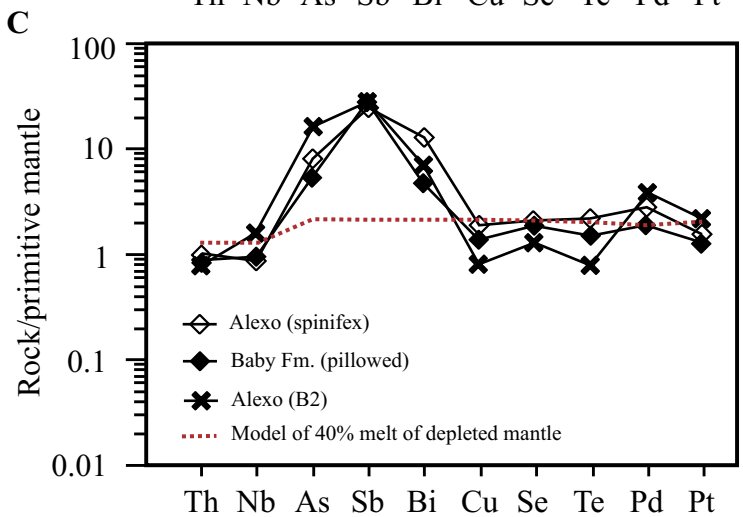
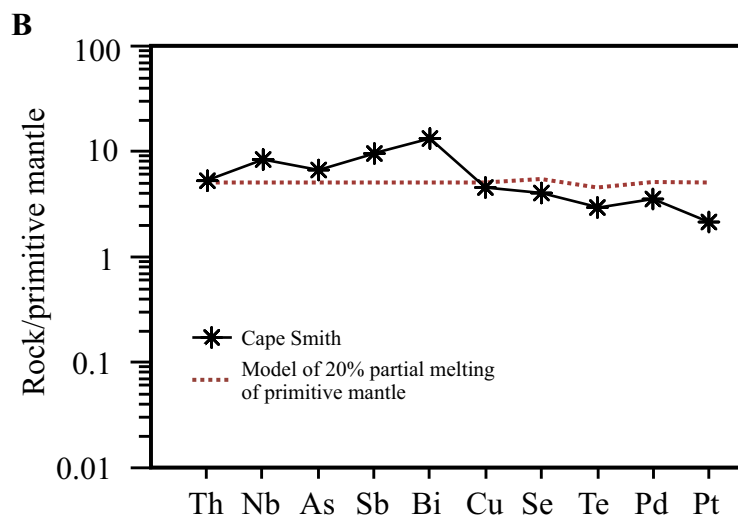
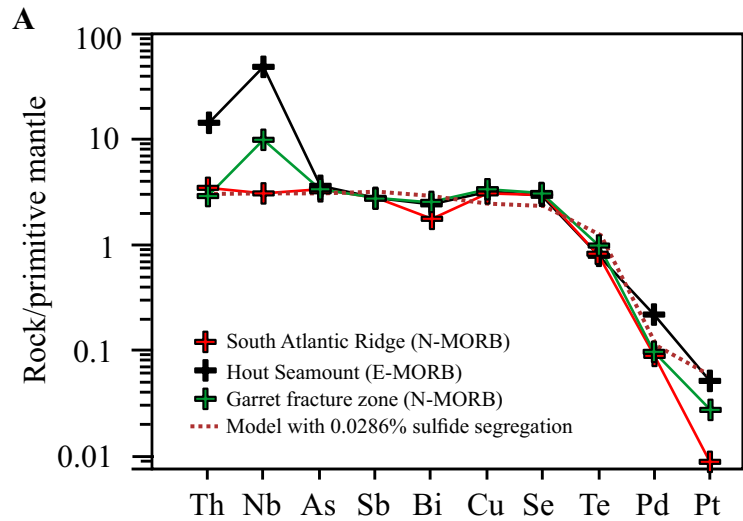


Fig. 2



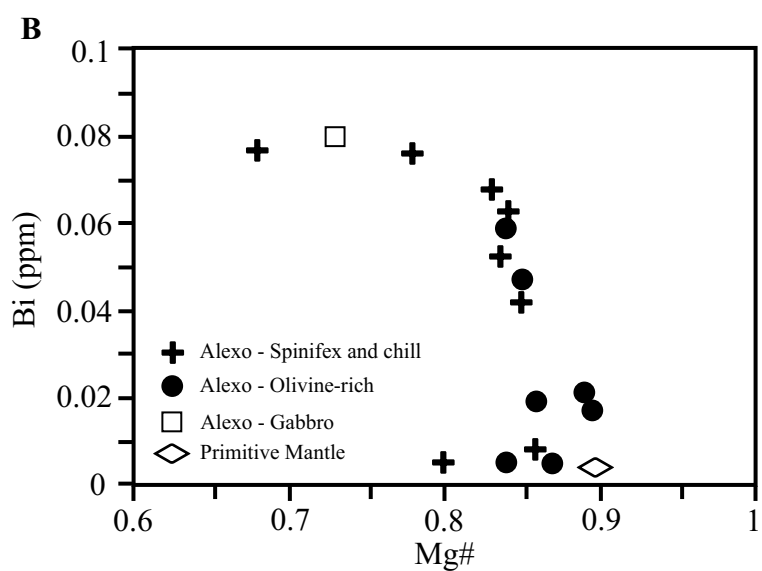
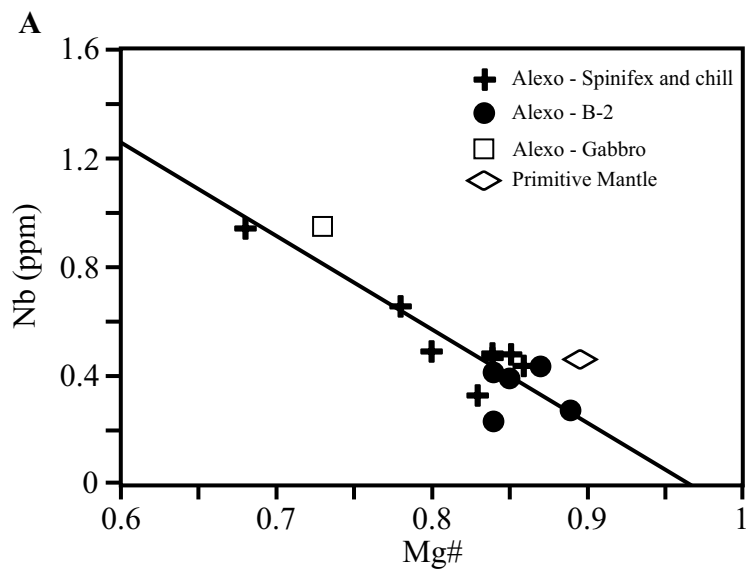


Fig. 4

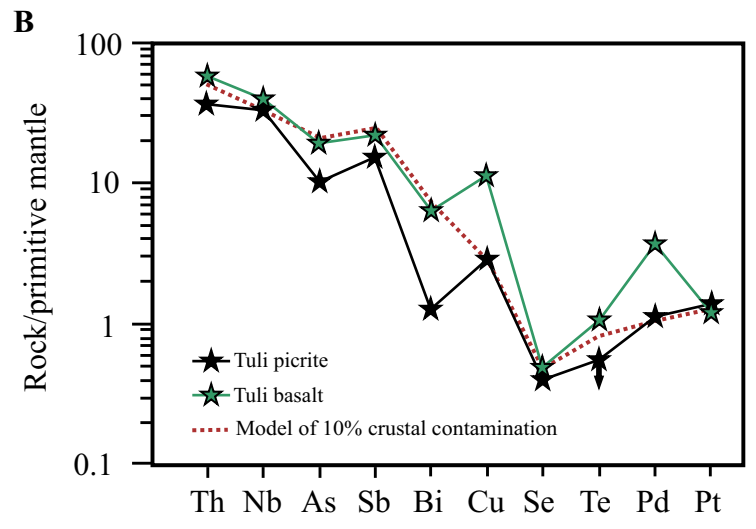
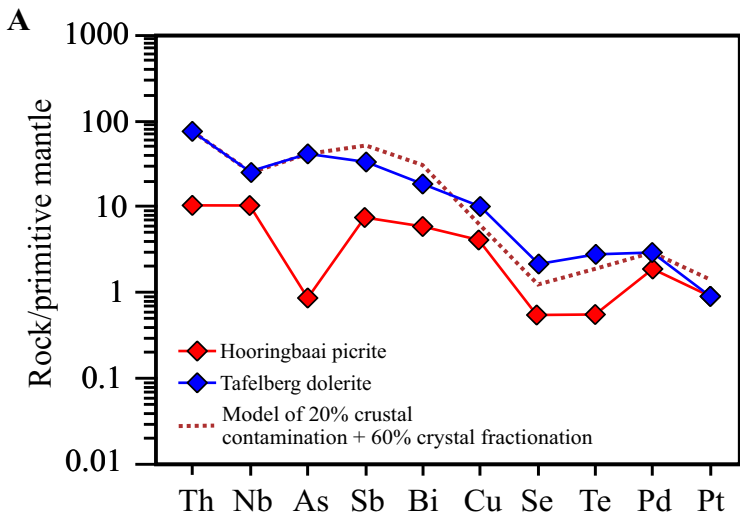


Fig. 5

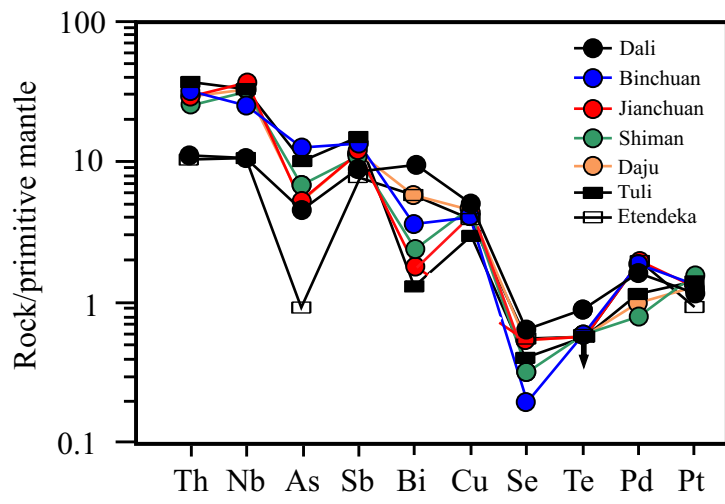
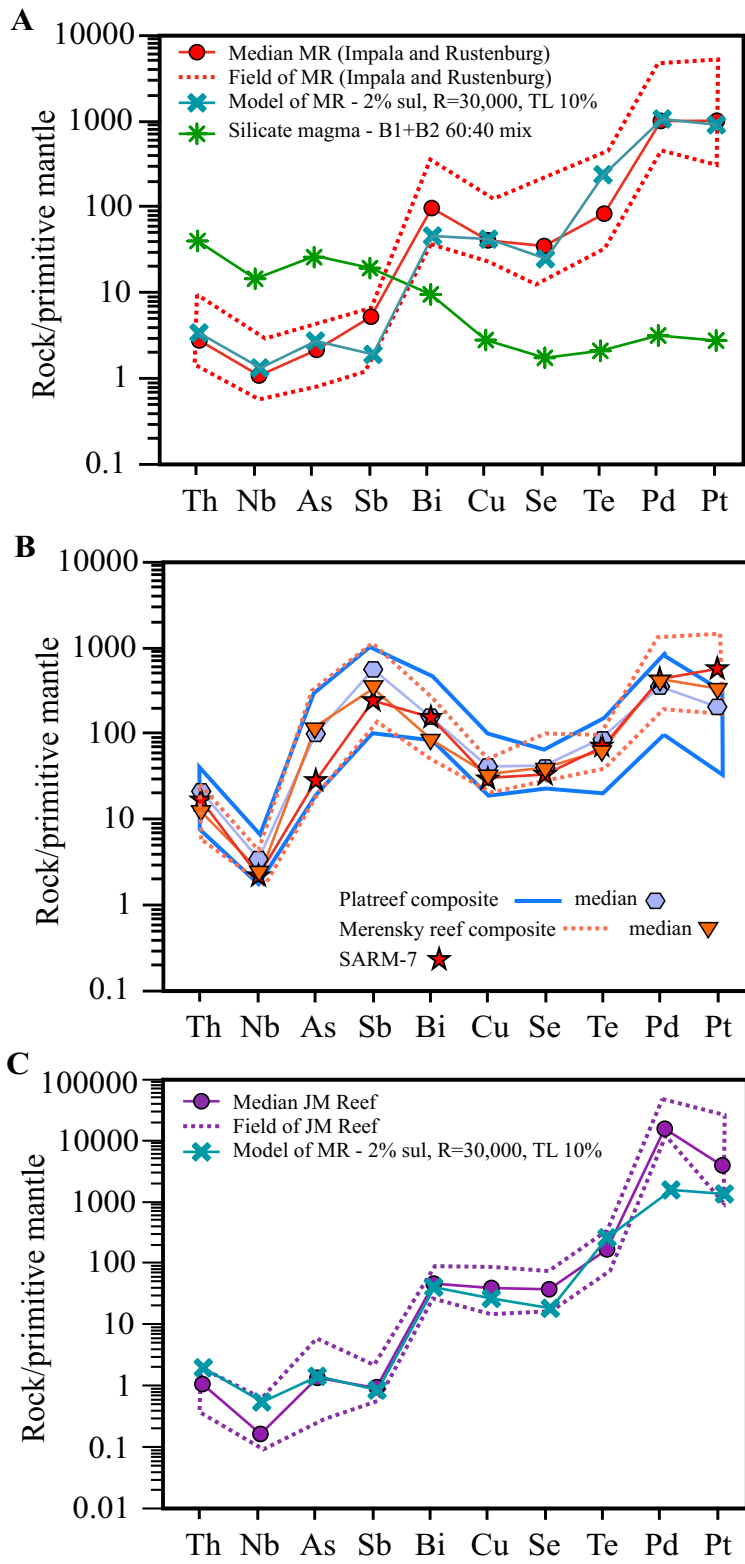


Fig. 6

Fig. 7



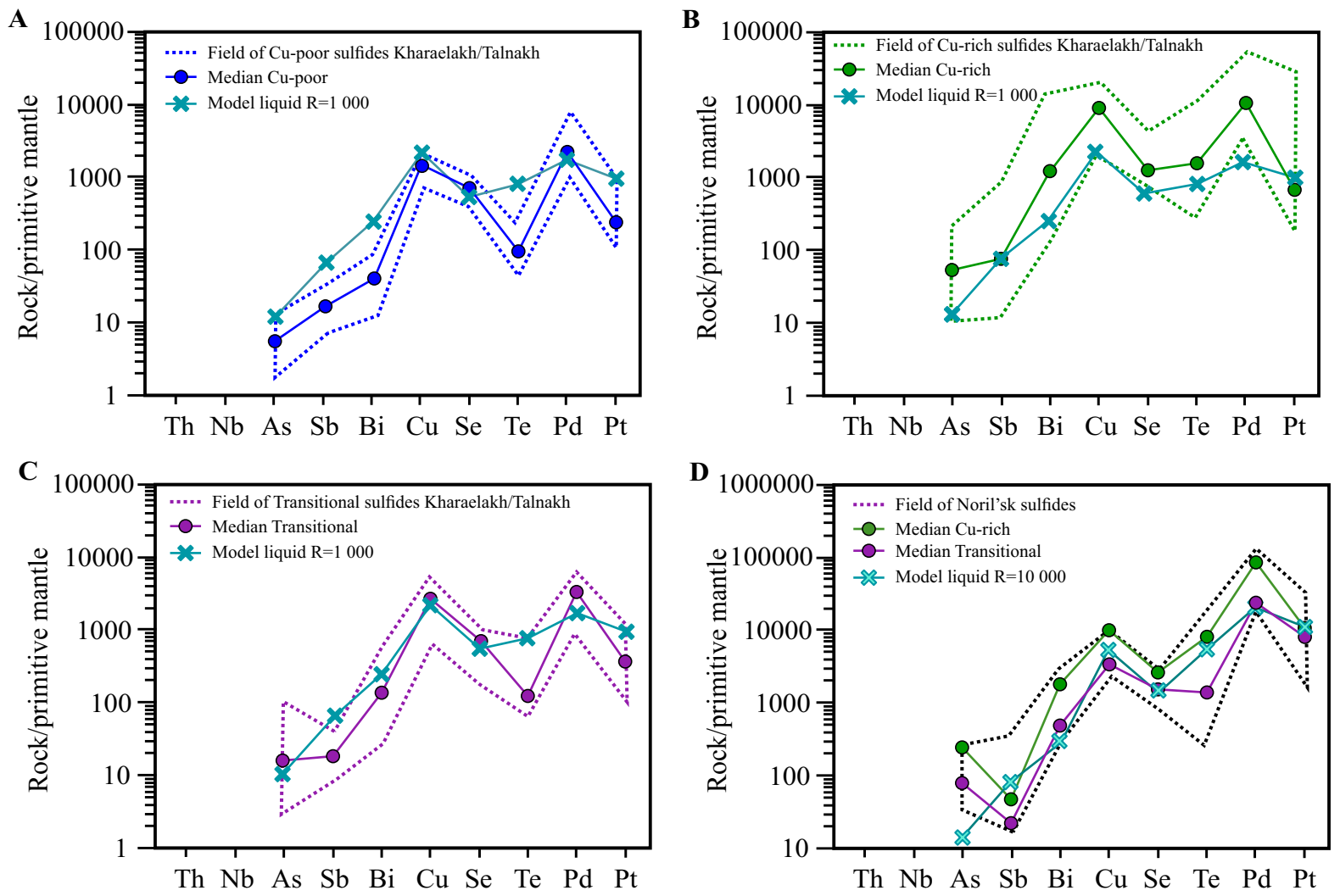


Fig. 8

Appendix 1 for Distribution of Te, As, Bi, Sb and Se in; MORB, Komatiites and in Picrites and Basalts from Large Igneous Provinces: Implications for the Formation of Magmatic Ni-Cu-PGE Deposits. Barnes and Mansur (2021) Econ. Geol.

Sample description and Analytical Methods

1. Description of the samples and their geological context

Mid-ocean ridge basalts

We selected 11 MORB samples originally from the repository at the Lamont Doherty Earth Observatory of Columbia University, New York. Six of these are from the South Atlantic Ridge, located between 25°41'S and 26°32'S, whereas the other five are from the East Pacific Rise, two from the Hotu seamount chain and three from the Garrett fracture zone (Fig. 1). The samples were dredged from depths ranging from 2579 to 3999 m, which minimizes S degassing (Mathez, 1976; Patten et al., 2012).

These samples were previously studied by Patten et al. (2012 and 2013). who investigated the textural relations in quenched sulfide droplets, and the partition coefficients of chalcophile elements between sulfide and silicate liquids. The authors argued that the composition of the sulfide droplets was in equilibrium with the silicate glass and that they formed during fractional crystallization of silicate melts in magma chambers beneath the mid-ocean ridges. The seamount samples are LREE enriched and could represent E-MORB, the other samples are LREE depleted and represent N-MORB. These samples are used to examine sulfide segregation.

Cape Smith Belt

The Cape Smith belt is a Proterozoic E-W trending greenstone belt located in northern Quebec, between the Archean Superior and Churchill provinces (Fig. 1). The samples come from the Chukotat Group which consists of olivine-, pyroxene-, and plagioclase-phyric basalts and are interpreted to represent opening of an ocean basin (Picard et al., 1990; Barnes and Picard, 1993; Leshner, 2007). The selected samples are olivine and pyroxene phyric pillow basalts from the cycles d and e of the Chukotat Group and were interpreted as the MORB-like basalts (Hynes and Francis, 1982; Picard et al., 1990). Previous work (Barnes and Picard, 1993) shows that they are not depleted in PGE and hence sulfide segregation or sulfide retention in the source has not occurred. The samples have been metamorphosed to greenschist facies (Picard et al., 1990). The samples should show the effects of partial melting and crystal fractionation without sulfide control, but they could have been altered during metamorphism.

Abitibi Greenstone Belt and the Baby Group

The Abitibi Greenstone Belt is located in the central part of the Superior province, Canada (Fig. 1), occupying an area of approximately 600 km x 300 km, but most of its komatiitic rocks are restricted to the south-west portion of the belt (Ayer et al., 2002; Houlé et al., 2012). The selected samples are Al-undepleted komatiites from the Alexo olivine and clinopyroxene flows and have been previously studied by Barnes et al. (1983), Barnes (1985) and Meric (2018).

The Baby Group is part of the Belleterre-Angliers greenstone belt just south of the Abitibi belt (Fig. 1; Dimroth et al., 1983; Barnes et al., 1993; Sawyer and Barnes, 1994). The selected samples were previously studied by Mainville (1994) and are pillowed Al-undepleted komatiites and komatiitic basalts, similar to those from the Abitibi belt.

Both the Alexo and Baby komatiites are undepleted in PGE and thus have not experienced sulfide segregation or sulfide retention in the mantle. The Alexo komatiites have experienced prehnite-pumpellyite facies metamorphism, and the Baby komatiites have experienced greenschist metamorphism and thus both could show some alteration due to metamorphism.

Barberton greenstone belt

The Barberton greenstone belt is a 100 km x 50 km northeast-trending belt of supracrustal igneous and sedimentary rocks located at the eastern border of the Kaapvaal Craton (Fig. 1). The belt represents one of the oldest preserved ultramafic-mafic Archean sequences on the Earth with 3.5 to 3.2 Ga (Zeh et al., 2013). The selected samples are from the ultramafic-mafic basal sequence of the belt, named Onverwatch Group. These comprise 2 samples from each of the Sandspruit, Komati and Hooggenoeg formations. All the samples are Al-depleted spinifex-textured komatiites and were previously studied by Maier et al. (2009), who investigated the PGE distribution in these rocks and concluded that they are depleted in PGE. Although this depletion could be attributed to sulfide segregation Maier et al. (2009) attributed this depletion to the source of the komatiites being depleted in PGE due to incomplete mixing of the later veneer in the primitive mantle. These rocks have experienced greenschist metamorphism and may show evidence of alteration.

Emeishan Large Igneous Province

The Emeishan Large Igneous Province (ELIP) is a continental flood basalt province located in the western margin of the Yangtze craton in SW China (Fig. 1). The province covers an area of $\sim 2.5 \times 10^5 \text{ km}^2$, with a total volume of $\sim 0.3 \times 10^6 \text{ km}^3$ (Ali et al., 2005; Kamenestky et al., 2012). Geochronology studies shows that the rocks were emplaced around 260 Ma, over an interval of a few million years (Ali et al., 2005; Liu and Zhu, 2009), which is coincident with the end-Guadalupian mass extinction (Zhou et al., 2002; Xu et al., 2008). Several studies interpret the ELIP

as having formed by a plume-derived magmatism (Song et al. 2008; Xu et al.2008; Xiao et al., 2004).

The lava sequence ranges in thickness from 200 to more than 5000 m (Shellnutt, 2013) and consists mostly of tholeiitic basalts, which represent more than 95% of the magma volume. Based mainly on their Ti/Y ratios, Xu et al. (2001) classified the tholeiitic basalts of the ELIP into low and high-Ti basalts. The low-Ti basalts have Ti/Y lower than 500, whereas the high-Ti basalts have Ti/Y greater than 500.

The selected samples comprise 15 high-Ti and 9 low-Ti picrite lavass, and 9 flood basalts collected in the Jianchuan, Lijiang (Shiman and Daju sections), Dali and Binchuan, areas, in the western part of the ELI and 3 samples comprising olivine-phyric subvolcanic rocks from the Ertan area. Arguin et al. (2016) previously studied this collection of samples and concluded chromite, olivine and platinum-group mineral crystallization depleted the magma in Os, Ir, Ru and Rh during crystal fractionation, but did not greatly affect Pd or Pt concentrations. These samples are not metamorphosed, but as they are subaerial they may have experienced degassing.

Etendeka Igneous Province

The Etendeka Province is part of the Large Igneous Province denominated Paraná-Etendeka Magmatic Province (PEMP; Peate, et al. 1997; Marsh et al., 2001). The generation of the PEMP is associated with the opening of the South Atlantic Ocean during the Early Cretaceous, around 132 Ma (Renne et al., 1996; Peate, et al. 1997). The Etendeka Province corresponds to the remnant of the associated magmatism in northwest Namibia (Fig. 1), with a present-day outcrop area of approximately 78,000 km².

The selected samples consist of 3 dolerite dykes and 3 picritic dykes from the Tafelberg and Horingbaai areas and have been previously studied by Maier et al. (2003), who investigated the distribution of PGE in southern African flood basalts. The Horingbaai dykes represent asthenospheric mantle derived magmas without clear crustal contamination, whereas crustal assimilation is suggested to be more important for Tafelberg dykes (Erlank et al., 1984; Peate and Hawkesworth, 1996; Thompson et al., 2001). Maier et al. (2003) argue that sulfide segregation was not significant during the formation of the rocks from the Etendeka Province. The rocks are not metamorphosed, but they are subvolcanic, and thus could show the effects of degassing and variable degrees of crustal contamination.

Karoo Igneous Province

The Karoo Large Igneous Province is located at the southern part of Africa (Fig. 1) and formed during the separation of the African and Antarctica portions of Gondwana, around 180 Ma (Marsh et al., 1997; Riley and Knight, 2001; McClintock et al., 2008). Most of the remnants of the Karoo province comprise sills intruding sediments of the Karoo Supergroup, and up to 1600 m thick lava pile preserved in the Lesotho region, and a volcanic sequence around the Lebombo–Sabi region (Eales et al., 1984; Duncan et al., 1997; Marsh et al. 1997). Radiometric results show that the emplacement of the lavas took place in a brief event of potentially less than 1 Ma (Jourdan et al., 2007; Svensen et al., 2012).

The samples comprise tholeiitic basalts from the Barkly East (n=8) and Lesotho (n=13) formations, in the southwestern portion of the province, and from the Tuli syncline (n=3) at the northern portion of the province. Strontium isotopic work indicates that the rocks have undergone crustal contamination (Marsh et al., 1997). Based on the PGE content of the lavas Maier et al.

(2003) argued that they have not undergone significant sulfide segregation. The rocks are not metamorphosed, and some are vesicular. They could show the combined effects of degassing and crustal contamination.

Siberian Flood Basalts

The Siberian Large Igneous Province, located across western and eastern Siberia (Fig. 1), comprises very large alkaline, mafic and felsic magmatism erupted at the Permo-Triassic boundary. Geochronology studies shows that the emplacement of the Siberian LIP lasted less than 1 Ma (between 252.3 and 251.3 Ma; Reichow et al., 2009; Burgess et al., 2014). The magmas interacted with the volatile-rich sedimentary rocks, and associated gases released during the Siberian LIP event are believed to be responsible for the end-Permian mass extinction (Polozov et al., 2016, and references therein).

The selected samples are from the Noril'sk region, in the northwest portion of the province, where the full volcanic sequence has been drilled (Fedorenko, 1994). The lower formations were derived from high-Ti magmas with plume characteristics, whereas the upper formations were derived from low-Ti magmas contaminated with continental crust (Lightfoot et al., 1993). The selected samples from the Nadezhdinsky (Nd), Morongovsky (Mr), Mokulaevsky (Mk) and Kharayelakhsky (Kh) formations are part of the upper formations. Additionally, a sample from a shale from the Medvezhy open pit mine was also included as an example of a potential contaminant. Lightfoot et al. (1993) and Brüggemann et al. (1993) have shown that the Nd and parts of the Mr formations are depleted in both PGE and Cu and are thought to have segregated a sulfide liquid, with the Nd formation showing the greater degree of contamination. The Nd samples then could show the combined effects of sulfide segregation, degassing and crustal contamination

whereas the Mr, Mk and Kh samples should show the effects of degassing and crustal contamination. None of the samples are metamorphosed.

2. Analytical Methods

Tellurium, As, Bi, Sb and Se analyses were carried out by Hydride Generation-Atomic Fluorescence Spectrometry (HG-AFS) following the technique described by Mansur et al. (2020b), at LabMaTer Université du Québec à Chicoutimi (UQAC). Approximately 0.4 g of sample were digested with 5 ml of aqua regia (1:3 HNO₃:HCl) in close-caped beaker at 70°C for 24 hours. The aliquot was allowed to cool and diluted to 25 ml prior to mixing with a reductant solution (0.7% NaBH₄ and 0.4%NaOH). The mixed solution was analysed by Hydride Generation-Atomic Fluorescence Spectrometry (HG-AFS), using a continuous flow PSA Millennium Excalibur 10.055 from PS Analytical. Six calibration solutions with concentrations of 0.1, 0.25, 0.5, 1, 2.5 and 5 ppb were prepared using standard solutions of each element (PlasmaCAL, SCP Science, Quebec, Canada). The calibration solutions were mixed with the reagent blank prior to measurement, in the same proportion as sample aliquots. Calibration solutions were measured at the beginning and the end of each sequence of analysis to monitor fluctuations of the fluorescence signal, which were not observed. International reference materials (CH-4 and TDB-1 from Natural Resources Canada and OKUM from IAGEO), and a blank were determined at the same time as the samples. The detection limits based on 3 σ of the blank are 0.005, 0.003, 0.005, 0.005 and 0.002 ppm for Te, As, Bi, Sb and Se, respectively. The results for the reference materials agree with working values (electronic supplementary materials, ESM 3 Table 1).

For most samples, whole-rock analyses were already available in the original publications. For samples where they were not, the samples were analyzed at LabMaTer, (UQAC).

Approximately 0.2g of sample, 1g of flux powder (98.5% LiBO₂ and 1.5% LiBr) and 0.2g of NH₄NO₃ were mixed in a platinum crucible and melted using a Claisse Fluxer. The mixture was first heated up to 800°C for 5 minutes, followed by 4 minutes at 1050°C and 20 rotations per minute, and finally 3 minutes at 1100°C and 35 rotations per minute. The melting product is cool down for 3 minutes and a glass disk was produced. The glass disks were mounted in an epoxy block and analysed by laser ablation-inductively coupled plasma-mass spectrometry (LA-ICP-MS). The LA-ICP-MS analyses of the glass disks were performed using an Excimer 193 nm RESOLUTION M-50 laser ablation system (Australian Scientific Instrument) equipped with a double volume cell S-155 (Laurin Technic) and coupled with an Agilent 7900 mass spectrometer. The LA-ICP-MS tuning parameters were a laser frequency of 10 Hz, a power of 5 mJ/pulse, a dwell time of 7.5 ms, a rastering speed of 5 µm/s, and a fluence of 3 J/cm² and a beam size of 80µm. The gas blank was measured for 30s before switching on the laser for at least 60s. The ablated material was carried into the ICP-MS by an Ar–He gas mix at a rate of 0.8–1 L/min for Ar and 350 mL/min for He, and 2mL/min of nitrogen was also added to the mixture. Data reduction was carried out using the Iolite package for Igor Pro software (Paton et al., 2011). The isotopes measured for each element are reported in ESM, Table 2. The certified reference materials OKUM, KPT-1 (IAG reference materials), WPR-1 (CANMET) and UB-N (CNRS-CRPG), were used to monitor the results. The results obtained for the reference materials agree with the working values (ESM 3, Table 2).

References

- Ali, J. R., Thompson, G. M., Zhou, M.-F., and Song, X., 2005, Emeishan large igneous province, SW China: *Lithos*, v. 79, p. 475-489.
- Arguin, J.-P., Pagé, P., Barnes, S.-J., Yu, S.-Y., and Song, X.-Y., 2016, The effect of chromite crystallization on the distribution of osmium, iridium, ruthenium and rhodium in picritic magmas:

- an example from the Emeishan Large Igneous Province, Southwestern China: *Journal of Petrology*, v. 57, p. 1019-1048.
- Ayer, J., Amelin, Y., Corfu, F., Kamo, S., Ketchum, J., Kwok, K., and Trowell, N., 2002, Evolution of the southern Abitibi greenstone belt based on U–Pb geochronology: autochthonous volcanic construction followed by plutonism, regional deformation and sedimentation: *Precambrian Research*, v. 115, p. 63-95.
- Barnes, S.-J., 1985, The petrography and geochemistry of komatiite flows from the Abitibi Greenstone Belt and a model for their formation: *Lithos*, v. 18, p. 241-270
- Barnes, S.-J., and Picard, C., 1993, The behaviour of platinum-group elements during partial melting, crystal fractionation, and sulphide segregation: an example from the Cape Smith Fold Belt, northern Quebec: *Geochimica et Cosmochimica Acta*, v. 57, p. 79-87.
- Barnes, S.-J., Gorton, M., and Naldrett, A., 1983, A comparative study of olivine and clinopyroxene spinifex flows from Alexo, Abitibi greenstone belt, Ontario, Canada: *Contributions to Mineralogy and Petrology*, v. 83, p. 293-308.
- Barnes, S.-J., Couture, J.-F., Sawyer, E. W., and Bouchaib, C., 1993, Nickel-copper occurrences in the Belleterre-Angliers Belt of the Pontiac Subprovince and the use of Cu-Pd ratios in interpreting platinum-group element distributions: *Economic Geology*, v. 88, p. 1402-1418.
- Burgess, S. D., Bowring, S., and Shen, S.-Z., 2014, High-precision timeline for Earth's most severe extinction: *Proceedings of the National Academy of Sciences*, v. 111, p. 3316-3321.
- Dimroth, E., Imreh, L., Goulet, N., and Rocheleau, M., 1983, Evolution of the south-central segment of the Archean Abitibi belt, Quebec. Part II: Tectonic evolution and geomechanical model: *Canadian Journal of Earth Sciences*, v. 20, p. 1355-1373.
- Duncan, R. A., Hooper, P., Rehacek, J., Marsh, J., and Duncan, A., 1997, The timing and duration of the Karoo igneous event, southern Gondwana: *Journal of Geophysical Research: Solid Earth*, v. 102, p. 18127-18138.
- Eales H.V., Marsh J.S. and Cox K.G. 1984, The Karoo igneous province: An introduction. *In* Erlank, A. J., ed., *Petrogenesis of the Volcanic Rocks of the Karoo Province*, Geol. Soc. South Africa Spec. Publ. 13, p. 1-26.
- Erlank, A., Marsh, J., Duncan, A., Miller, R. M., Hawkesworth, C., Betton, P., and Rex, D., 1984, Geochemistry and petrogenesis of the Etendeka volcanic rocks from SWA Namibia. *In* Erlank, A. J., ed., *Petrogenesis of the Volcanic Rocks of the Karoo Province*, Geol. Soc. South Africa Spec. Publ. 13, p. 195-146.

- Fedorenko, V.A. 1994, Evolution of magmatism as reflected in the volcanic sequence of the Noril'sk region. *in* Lightfoot, P. C., and Naldrett, A. J., eds., Proceedings of the Sudbury-Noril'sk symposium, Special Publication no. 5, Ontario Geological Survey p. 171-184.
- Houlé, M. G., Lesher, C. M., and Davis, P. C., 2012, Thermomechanical erosion at the Alexo Mine, Abitibi greenstone belt, Ontario: implications for the genesis of komatiite-associated Ni-Cu-(PGE) mineralization: *Mineralium Deposita*, v. 47, p. 105-128.
- Hynes, A., and Francis, D. M., 1982, A transect of the early Proterozoic Cape Smith foldbelt, New Quebec: *Tectonophysics*, v. 88, p. 23-59
- Jourdan, F., Bertrand, H., Schärer, U., Blichert-Toft, J., Féraud, G., and Kampunzu, A., 2007, Major and trace element and Sr, Nd, Hf, and Pb isotope compositions of the Karoo large igneous province, Botswana-Zimbabwe: lithosphere vs mantle plume contribution: *Journal of Petrology*, v. 48, p. 1043-1077.
- Kamenetsky, V. S., Chung, S.-L., Kamenetsky, M. B., and Kuzmin, D. V., 2012, Picrites from the Emeishan Large Igneous Province, SW China: a compositional continuum in primitive magmas and their respective mantle sources: *Journal of Petrology*, v. 53, p. 2095-2113.
- Lesher, C.M., 2007, Ni-Cu-(PGE) Deposits in the Raglan Area, Cape Smith Belt, New Québec, *in* Goodfellow, W. D., ed., *Mineral Deposits of Canada: A Synthesis of Major Deposit-Types, District Metallogeny, the Evolution of Geological Provinces, and Exploration Methods: Special Publication No. 5*, Mineral Deposits Division, Geological Association of Canada, p. 351-386.
- Lightfoot, P., Hawkesworth, C., Hergt, J., Naldrett, A., Gorbachev, N., Fedorenko, V., and Doherty, W., 1993, Remobilisation of the continental lithosphere by a mantle plume: major-, trace-element, and Sr-, Nd-, and Pb-isotope evidence from picritic and tholeiitic lavas of the Noril'sk District, Siberian Trap, Russia: *Contributions to Mineralogy and Petrology*, v. 114, p. 171-188.
- Liu C. and Zhu R., 2009, Geodynamic significances of the Emeishan basalts. *Frontiers in Earth Science*, v. 16, p. 52-69.
- Maier, W. D., Barnes, S.-J., and Marsh, J. S., 2003, The concentrations of the noble metals in Southern African flood-type basalts and MORB: implications for petrogenesis and magmatic sulphide exploration: *Contributions to Mineralogy and Petrology*, v. 146, p. 44-61
- Maier, W. D., Barnes, S. J., Campbell, I. H., Fiorentini, M. L., Peltonen, P., Barnes, S.-J., and Smithies, R. H., 2009, Progressive mixing of meteoritic veneer into the early Earth's deep mantle: *Nature*, v. 460, p. 620-623.
- Mainville, M., 1994, Les komatiites et tholeiites a la base du Groupe de Baby, Temiscamingue : Unpub MSc, Université du Québec à Chicoutimi.

- Mansur, E. T., Barnes, S. J., Savard, D., and Webb, P. C., 2020, Determination of Te, As, Bi, Sb and Se (TABS) in Geological Reference Materials and GeoPT Proficiency Test Materials by Hydride Generation-Atomic Fluorescence Spectrometry (HG-AFS): *Geostandards and Geoanalytical Research*, v. 44, p. 147-167.
- Marsh, J., Ewart, A., Milner, S., Duncan, A., and Miller, R. M., 2001, The Etendeka Igneous Province: magma types and their stratigraphic distribution with implications for the evolution of the Paraná-Etendeka flood basalt province: *Bulletin of Volcanology*, v. 62, p. 464-486.
- Mathez, E.A., 1976, Sulfur solubility and magmatic sulfides in submarine basalt glass: *Journal of Geophysical Research*, v. 81, p. 4269-4276.
- McClintock, M., White, J.D.L., Houghton, B.F. and Skilling, I.P., 2008, Physical volcanology of a large crater-complex formed during the initial stages of Karoo flood basalt volcanism, Sterkspruit, Eastern Cape, South Africa: *Journal of Volcanology and Geothermal Research*, v. 172(1-2), p. 93-111.
- Meric, J., 2018, Le ruthénium (Ru), iridium (Ir), osmium (Os) et rhodium (Rh) et les éléments traces dans des chromites de komatiites issues de la zone Alexo et de la zone Hart, (Abitibi, Ontario): un outil diagnostique pour l'exploration de systèmes fertiles: Unpub MSc, Université du Québec à Chicoutimi.
- Paton, C., Hellstrom, J., Paul, B., Woodhead, J., and Hergt, J. (2011) Iolite: Freeware for the visualisation and processing of mass spectrometric data: *Journal of Analytical Atomic Spectrometry*, v. 26, p. 2508-2518.
- Patten, C., Barnes, S-J., and Mathez, E.A., 2012, Textural variations in MORB sulfide droplets due to differences in crystallization history: *The Canadian Mineralogist*, v. 50(3), p. 675-692.
- Patten, C., Barnes, S-J., Mathez, E.A., and Jenner, F.E., 2013, Partition coefficients of chalcophile elements between sulfide and silicate melts and the early crystallization history of sulfide liquid: LA-ICP-MS analysis of MORB sulfide droplets: *Chemical Geology*, v. 358, p. 170-188.
- Peate, D., and Hawkesworth, C.J., 1996, Lithospheric to asthenospheric transition in low-Ti flood basalts from southern Parana, Brazil: *Chemical Geology*, v. 127, 1-24.
- Peate, D. W., 1997, The Parana-Etendeka Province. *in* Mahoney, J.J. and Coffin, M.F. (eds.) *Large Igneous Provinces: Continental, Oceanic, and Planetary Flood Volcanism*: American Geophysical Union, Monograph 100, p. 217-245.
- Polozov, A.G., Svensen, H.H., Planke, S., Grishina, S.N., Fristad, K.E., and Jerram, D.A., 2016, The basalt pipes of the Tunguska Basin (Siberia, Russia): High temperature processes and volatile degassing into the end-Permian atmosphere: *Palaeogeography Palaeoclimatology Palaeoecology*, v. 441, p. 51-64.

- Reichow, M.K., Pringle, M.S., Al'Mukhamedov, A.I., Allen, M.B., Andreichev, V.L., Buslov, M.M., Davies, C.E., Fedoseev, G.S., Fitton, J.G., Inger, S., Medvedev, A.Y., Mitchell, C., Puchkov, V.N., Safonova, I.Y., Scott, R.A., and Saunders, A.D., 2009, The timing and extent of the eruption of the Siberian traps large igneous province: implications for the end-Permian environmental crisis: *Earth and Planetary Science Letters*, v. 277, p. 9-20.
- Renne, P.R., Glen, J.M., Milner, S.C., and Duncan, A.R., 1996, Age of Etendeka volcanism and associated intrusions in southwestern Africa: *Geology*, v. 24, p. 659-662.
- Riley, T.R., and Knight, K.B., 2001, Age of pre-break-up Gondwana magmatism: *Antarctic Science*, v. 13(2), p. 99-110.
- Sawyer, E.W., and Barnes, S.-J., 1994, Thrusting, magmatic intraplate, and metamorphic core complex development in the Archaean Belleterre-Angliers greenstone belt, Superior Province, Quebec, Canada: *Precambrian Research*, v. 68(3-4), p. 183-200.
- Shellnutt, J.G., 2013, The Emeishan large igneous province: a synthesis: *Geoscience Frontiers*, v. 5, p. 369-394.
- Song, X.Y., Qi H.W., Robinson, P.T., Zhou, M.F., Cao, Z.M., and Chen, L.M., 2008, Melting of the subcontinental lithospheric mantle by the Emeishan mantle plume; evidence from the basal alkaline basalts in Dongchuan, Yunnan, Southwestern China: *Lithos*, v. 100, p. 93-111.
- Svensen, H., Corfu, F., Polteau, S., Hammer, Ø., and Planke, S., 2012, Rapid magma emplacement in the Karoo large igneous province: *Earth and Planetary Science Letters*, v. 325, p. 1-9.
- Thompson, R.N., Gibson, S.A., Dickin, A.P., and Smith, P.M., 2001, Early Cretaceous basalt and picrite dykes of the southern Etendeka region, NW Namibia: windows into the role of the Tristan mantle plume in Parana-Etendeka magmatism: *Journal of Petrology*, v. 42, p. 2049-2081.
- Xiao, L., Xu, Y.G., Mei, H.J., Zheng, Y.F., He, B., and Pirajno, F., 2004, Distinct mantle sources of low-Ti and high-Ti basalts from the western Emeishan large igneous province, SW China: implications for plume-lithosphere interaction: *Earth and Planetary Science Letters*, v. 228, p. 525-546.
- Xu, Y., Chung, S. L., Jahn, B. M., and Wu, G., 2001, Petrologic and geochemical constraints on the petrogenesis of Permian–Triassic Emeishan flood basalts in southwestern China: *Lithos*, v. 58, p. 145-168.
- Xu, Y.G., Luo, Z.Y., Huang, X.L., He, B., Xiao, L., Xie, L.W., and Shi, Y.R., 2008, Zircon U-Pb and Hf isotope constraints on crustal melting associated with the Emeishan mantle plume: *Geochimica et Cosmochimica Acta*, v. 72, p. 3084-3104.
- Zeh, A., Gerdes, A., and Heubeck, C., 2013, U–Pb and Hf isotope data of detrital zircons from the Barberton Greenstone Belt: constraints on provenance and Archaean crustal evolution: *Journal of the Geological Society of London*, v. 170(1), p. 215-223.

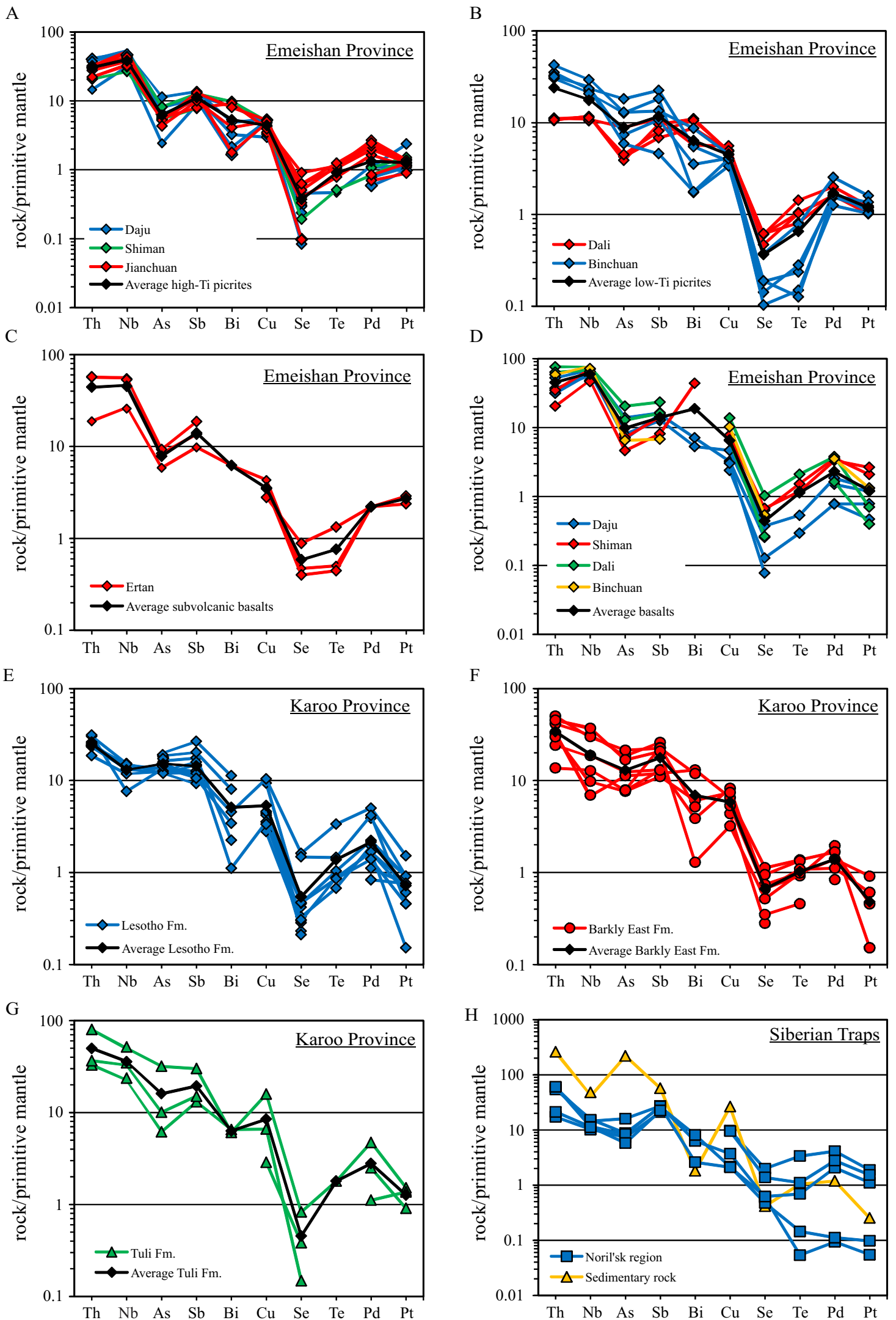
Zhou, M.F., Malpas, J., Song, X.Y., Robinson, P.T., Sun, M., Kennedy A.K., and Keays R.R., 2002, A temporal link between the Emeishan large igneous province (SW China) and the end-Guadalupian mass extinction. *Earth and Planetary Science Letters*, v. 196(3-4), p. 113-122.

READ ME

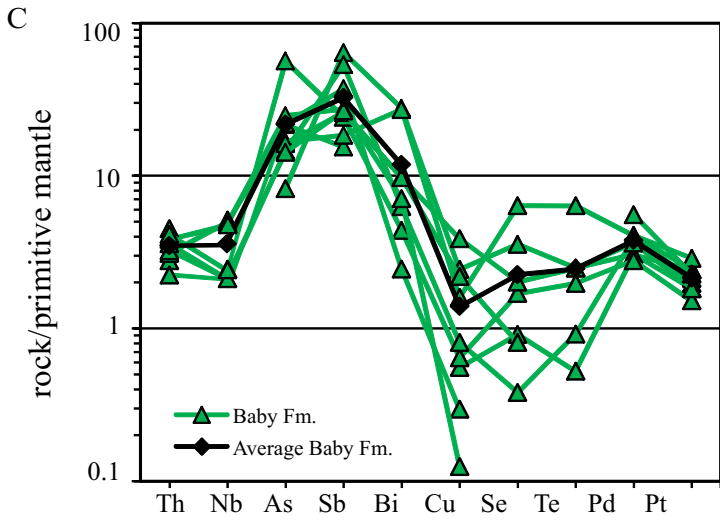
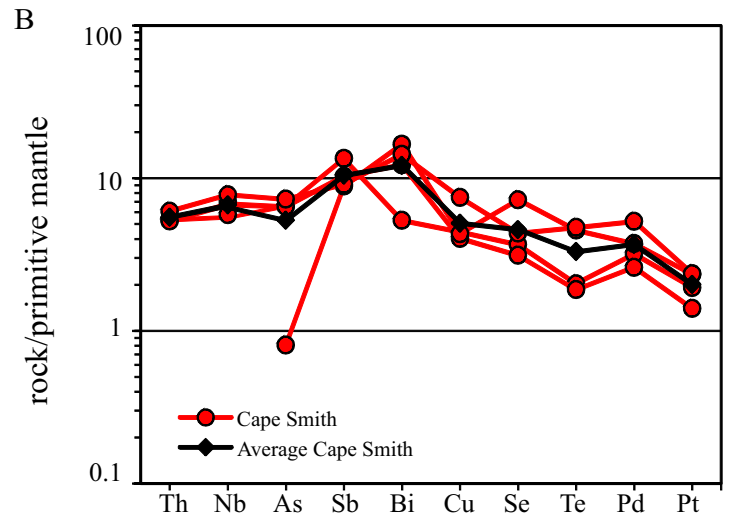
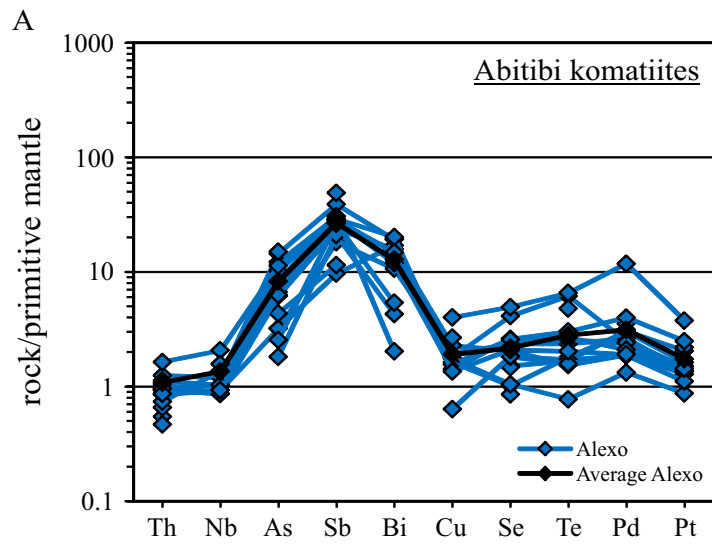
Online Resource (ESM) for
**Distribution of Te, As, Bi, Sb and Se (TARSe) in MORB, Komatiites and in Picrites and Basalts from Large
Igneous Provinces: Implications for the Formation of Magmatic Ni-Cu-PGE Deposits**
* Sarah-Jane Barnes¹, Eduardo T. Manau¹
Economic Geology

¹Sciences de la Terre, Université du Québec à Chicoutimi, Québec, G7H 2B1, Canada

*Corresponding author - E-mail address: sjbarnes@uqac.ca



Appendix 3 Fig A1 A high-Ti picrites – Emeishan Province; B low-Ti picrites – Emeishan Province; C Subvolcanic sill – Emeishan Province; D basalts – Emeishan Province; E Lesotho Formation – Karoo Province; F Barkly East Formation – Karoo Province; G Tuli Formation – Karoo Province; H Siberia



Appendix 3 Barnes and Mansure (2021) Econ Geol
 Figure A2 – Primitive mantle normalized Th, Nb, Cu, TABS, Pd
 and Pt plots: A Alexo – Abitibi Greenstone belt B Cape Smith
 belt; C Baby Formation; Primitive mantle values from
 Lyubetskaya and Korenaga (2007)